

USOS DEL SUBJUNTIVO

DESEO, SENTIMIENTO, VOLUNTAD

español online

VERBOS: Gustar (y verbos con la misma estructura: encantar, dar igual, importar, aburrir...), odiar, preferir, esperar, soportar - aguantar, querer, desear, tener ganas de, alegrarse de, estar harto-a de...

ESTRUCTURAS:

OJALA
QUE

+ SUBJUNTIVO

La regla es:



MISMO SUJETO

DIFERENTE SUJETO



INFINITIVO

SUBJUNTIVO

Ojalá no llueva en la boda
¡que tengas un finde genial!

Me encanta viajar

Me encanta que tú viajes



CONSEJO, MANDATO, RUEGO

VERBOS: Decir (cuando significa ordenar, mandar), comentar, pedir, rogar, suplicar, recomendar, aconsejar, ordenar, mandar, repetir, permitir, prohibir, exigir...

Do it!

Te suplico que limpies tu habitación

Te recomiendo que visites el parque de El Retiro en Madrid



OPINIÓN NEGATIVA

VERBOS: Creer, pensar, opinar, decir, estar convencido-a de, suponer, sostener, mantener, sospechar, imaginar, considerar...

La regla es:

OPINIÓN AFIRMATIVA

OPINIÓN NEGATIVA



INDICATIVO

SUBJUNTIVO

Creo que tiene una casa muy bonita

No creo que tenga una casa muy bonita



¡CUIDADO! *¡Es afirmativa!*
Creo que...no tiene una casa muy bonita



FUTURO

Cuando situamos la acción en el futuro y usamos:

Cuando, en cuanto, apenas, tan pronto como, a medida que, cada vez que, siempre que, mientras, desde / hasta que (no), antes/ después de que...

En cuanto llegues a casa, llámame

Cuando tenga hijos les enseñaré a hablar español



FINALIDAD

Para, a fin de, con el fin de, con la idea de, con la intención de...

La regla es:



MISMO SUJETO

DIFERENTE SUJETO



INFINITIVO

SUBJUNTIVO

Estudio español para comunicarme con mis amigos españoles
Estudio español para que mis amigos españoles me entiendan



DUDA Y PROBABILIDAD

tal vez, quizá(s),
probablemente,
posiblemente,
seguramente

a lo mejor,
igual,
lo mismo,
seguro que



es (im)posible que
es (im)probable que
Puede ser que
puede que

INDICATIVO O SUBJUNTIVO

Miguel quizá está/ esté
en casa

INDICATIVO

Lo mismo está en casa

SUBJUNTIVO

Puede que esté en casa

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Genaro Borrego,M. Asensi,E. Prieto,1985 **The Spanish Subjunctive: A Reference for Teachers** Hans-Jorg

Busch,2017-03-27 A Complete Guide to the Spanish Subjunctive is the most complete reference guide to the use of the subjunctive in Spanish Along with an exhaustive review of published literature on the subjunctive the book also includes a thorough discussion of the uses and meanings of the subjunctive as well as examples throughout drawn from linguistic corpora such as the CREA database The book presents a comprehensive theory of the subjunctive and provides practical rules for understanding teaching and acquiring the Spanish subjunctive This book includes Your Turn sections that invite readers to reflect on the content discussed and on their own experiences in teaching the subjunctive A Synopsis section that summarizes the content of the work and offers practical suggestions for teaching the subjunctive Two indexes providing a summary of verb conjugation in the subjunctive and an alphabetical list of expressions used with the subjunctive

Cuestiones del español como lengua extranjera Emma Martinell Gifre,Mar Cruz Piñol,2007-01-16 **New**
Directions in Hispanic Linguistics Alejandro Cortazar,2014-04-11 This volume addresses some lacunae in Hispanic linguistic research by focusing on new scholarly directions exploring understudied topics as well as speech communities and presenting new takes on relevant linguistic and sociocultural issues This publication answers questions which have emerged as a result of the rapid increase in Hispanic linguistic research since the latter part of the twentieth century or that have remained open in spite of it With the rapid growth of Hispanic Linguistics during the 21st century the topics included in this volume are representative of the breadth vitality and interdisciplinarity of contemporary linguistic scholarship They also reflect that linguistics in general has become more methodologically sophisticated This book is comprised of twelve chapters divided into three parts Part I addresses language ideology and language contact issues that are embedded in important sociolinguistic and cultural topics chronologically spanning from the 16th century to the present Although these issues take place in Spain the United States Turkey and Ecuador they pertain ideologically to all corners of the Hispanic World and beyond Part II is devoted to pragmatics and language variation with topics that transport us to Colombia Mexico Spain and Venezuela The study of politeness strategies shows how Spanish speakers reduce social distance between interlocutors as they make conversation a pleasant and cooperative meeting place Concurrently sociolinguistic innovations reveal interesting parallels among several speech communities Part III explores linguistic variation as it relates to theoretical structural and instructional issues Although these topics are analyzed based mainly on linguistic usage in Bolivia Colombia the Dominican Republic Ecuador Panama and Spain as with the rest of this volume their relevance reaches far beyond the confines of the Hispanic World This book is unique in multiple ways and complements a number of existing publications

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the

titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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 $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
 1 mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction $\text{NaOH (aq)} + \text{C}_6\text{H}_4\text{COOH (aq)} \rightarrow \text{C}_6\text{H}_4\text{COONa (aq)} + \text{H}_2\text{O (l)}$ the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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