Solute-Solvent Interactions

- Solubility varies greatly with the type of compounds involved.
- "Like dissolves like" is a rough but useful rule for predicting whether one substance will dissolve in another.
- What makes substances similar depends on: -
 - type of bonding
 - polarity or nonpolarity of molecules
 - intermolecular forces between the solute and solvent











Solute Solvent Interactions

Johannes F. Coetzee, Calvin D. Ritchie



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