

Name: _____ Date: _____ Class: _____

- 1 What kind of **reaction** has occurred in this picture?

A synthesis
B decomposition
C single replacement
D double replacement



- 2 In order for a **chemical change** to take place, _____.

A products must turn into reactants
B reactants must turn into products
C heat must be given off
D heat must be added



- 3 What **type** of **chemical reaction** is shown in the diagram below?



A double replacement
B synthesis
C decomposition
D single replacement

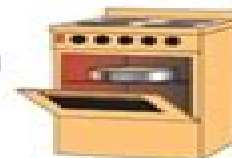
- 4 What is the **evidence** in this picture that a **chemical reaction** has occurred?

A light and heat
B sound
C change of shape
D change in mass



- 5 Baking a cake is an endothermic reaction that chemically changes the batter into a cake. For an **endothermic reaction** to occur, _____.

A energy must be released
B energy must be used
C shaking is necessary
D no reactants are used



- 6 Which statement is correct for the **reaction** shown below?

A it is a decomposition reaction
B it is a synthesis reaction to make hydrochloric acid
C mixing certain acids and bases makes salt and water
D only a physical change has occurred



- 7 What is one way of **speeding up** any chemical reaction?

A take away the products
B reduce the amount of reactants
C increase the amount of reactants
D decrease the temperature



- 8 What **type** of reaction is the chemical reaction shown below?



A double replacement
B single replacement
C synthesis
D decomposition

Teachers Manual The Chemical World Activities And Explorations

Dennis Coon



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