



The Havana Room

Jessica Page Morrell



The Havana Room:

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serpentine subgroup part of the kaolinite serpentine group in the category of phyllosilicates are greenish brownish or spotted minerals commonly found in serpentinite they are used as a source of magnesium and asbestos and as decorative stone

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serpentinization is a hydration and metamorphic transformation of ferromagnesian minerals such as olivine and pyroxene in mafic and ultramafic rock to produce serpentinite 1 minerals formed by serpentinization include the serpentine group minerals antigorite lizardite chrysotile brucite talc ni fe alloys and magnetite

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of characteristic of or resembling a serpent as in form or movement having a winding course as a road sinuous shrewd wily or cunning noun a device on a harquebus lock for holding the

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serpentinite is a rock composed predominantly of one or more serpentine group minerals the name originating from the similarity of the texture of the rock to that of the skin of a snake serpentinite has been called serpentine or serpentine rock particularly in older geological texts and in wider cultural settings

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serpentine any of a group of hydrous magnesium rich silicate minerals the composition of these common rock forming minerals approximates $\text{mg}_3\text{si}_2\text{o}_5\text{oh}_4$ serpentine generally occurs in three polymorphs chrysotile a fibrous variety used as asbestos antigorite a variety occurring in either

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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below
$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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aq k aq cl aq agcl s

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web the net ionic equation for equation 1 is given as equation 3 $\text{s}^{2+}(\text{aq}) + \text{Fe}^{2+}(\text{aq}) \rightarrow \text{FeS}(\text{s})$ eq 3 8 note that the phase indicator aq is optional on ionic species but is

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