

# ORGANIC CHEMISTRY

## FUNDAMENTALS

### Types of Organic Compounds

**Organic chemistry** is the study of natural and synthetic materials that have carbon atoms as the key chemical feature. There are more than one million known organic compounds.

#### HYDROCARBON

**ALKANE**  
 $\text{C}-\text{H}$   
 $\text{C}-\text{C}$   
 • Ethane:  $\text{C}_2\text{H}_6$   
 • Methyl (Me):  $-\text{CH}_3$   
 • Ethyl (Ety):  $-\text{C}_2\text{H}_5$

#### ALKENE

• Ethene:  $\text{C}_2\text{H}_4$   
 • Diene: Two  $\text{C}=\text{C}$   
 • Triene: Three  $\text{C}=\text{C}$

#### ALKYNE

• Ethyne:  $\text{C}_2\text{H}_2$

#### AROMATIC

• Benzene:  $\text{C}_6\text{H}_6$   
 • Aroclor:  $\text{C}_6\text{H}_5$  (Ar)

#### -O- ADDED

##### ALCOHOL

• Methanol:  $\text{Me}-\text{OH}$  (methyl alcohol)  
 • Phenol:  $\text{Ar}-\text{OH}$   
 • Diol/Glycol: (2  $-\text{OH}$ )  
 • Glycerol: (3  $-\text{OH}$ )

##### ETHER

• Ethoxyethane:  $\text{Et}-\text{O}-\text{Et}$  (diethyl ether)

##### EPOXY

• Oxirane:  $\text{C}_2\text{H}_4\text{O}$  (ethylene oxide)

##### PEROXIDE

• Acetone Peroxide:  $\text{C}_4\text{H}_8\text{O}_5$

• Benzoyl Peroxide:  $\text{C}_{14}\text{H}_{18}\text{O}_5$

#### >C=O ADDED

##### ALDEHYDE

• Methanal:  $\text{H}_2\text{CO}$  (formaldehyde)  
 • Benzaldehyde:  $\text{Ar}-\text{CHO}$

##### KETONE

• 2-propanone:  $\text{Me}-\text{CO}-\text{Me}$  (dimethyl ketone, acetone)  
 • Diketone:  $\text{R}-\text{CO}-\text{R}'-\text{CO}-\text{R}''$

#### >COO ADDED

##### CARBOXYLIC ACID

• Ethanoic Acid:  $\text{Me}-\text{COOH}$  (acetic acid)  
 • Acetate Ion:  $\text{Me}-\text{COO}^-$   
 • Benzoic Acid:  $\text{Ar}-\text{COOH}$

##### Dicarboxylic Acid

$\text{HOOC}-\text{R}-\text{COOH}$

#### ESTER

• Ethyl Acetate:  $\text{Me}-\text{CO}-\text{OEt}$   
 Other derivatives:  
 • Peroxyacid:  $\text{R}-\text{CO}-\text{OOH}$   
 • Acid Anhydrides:  $\text{RCO}-\text{O}-\text{CO}-\text{R}'$

#### NITROGEN ADDED

##### AMINE

• Methyl Amine:  $\text{H}_3\text{C}-\text{NH}_2$   
 • Phenyl amine:  $\text{Ar}-\text{NH}_2$  (aniline)  
 •  $\text{R}-\text{NH}_2$  (1°),  $\text{RR}'\text{NH}$  (2°),  $\text{RR}'\text{R}''\text{N}$  (3°)

##### NITRO

$\text{R}-\text{NO}_2$

##### DIAZO

$\text{R}-\text{N}=\text{N}$

##### NITRILE

$\text{R}-\text{C}\equiv\text{N}$

• Methane Nitrile:  $\text{Me}-\text{CN}$

#### AMIDE

• Acetamide:  $\text{Me}-\text{CO}-\text{NH}_2$

#### SULFUR ADDED

• Thiol:  $\text{R}-\text{SH}$   
 • Thioether:  $\text{R}-\text{S}-\text{R}'$   
 • Disulfide:  $\text{R}-\text{S}-\text{S}-\text{R}'$   
 • Thiol Ester:  $\text{R}-\text{CO}-\text{SR}'$   
 • Sulfide:  $\text{R}-\text{SO}-\text{R}'$   
 • Sulfone:  $\text{R}-\text{SO}_2-\text{R}'$   
 • Sulfonic Acids:  $\text{R}-\text{SO}_3\text{H}$

#### HALOGEN ADDED

• Haloalkane:  $\text{Me}-\text{Cl}$  chloromethane  
 • Halohydrate:  $\text{Ar}-\text{X}$   
 • Chloroalkane:  $\text{Ar}-\text{Cl}$   
 • Acyl Halide:  $\text{R}-\text{CO}-\text{X}$   
 • Aryl Halide:  $\text{Ar}-\text{X}$

### Formulas & Isomers

- Molecular Formula:** Elemental symbols with subscripts denote the composition of a compound
- Empirical Formula:** Subscripts denote the relative elemental composition
- Graphical depiction**
  - Dash Formula:** Diagram all atoms, show bonds as dashes
  - Bond-Line Formula:** Hide H; carbon atoms are depicted as lines, but other atoms are shown explicitly



- Newman Projection:** 2-d depiction
- 3-dimensional:** Wedges of same bond denote structure

#### SUBDIVISIONS OF ISOMERS



- Constitutional Isomers:** Different bonding connectivity (e.g., rings, bonds, branching, substituent positions)
- Tautomers:** Easily interconverted structural isomers (e.g., keto-enol for ketones)
- Chiral:** Not identical with its mirror image
- Achiral:** Has a plane of symmetry (i.e., superimposable on its mirror image)
- Epimers:** Pair of diastereomers which differ only in the configuration of one atom
- More than 1 chiral center
  - n chiral centers,  $\leq 2^n$  stereoisomers
  - Meso: 2 chiral centers, 4 isomers: 3 stereoisomers, 1 achiral (mirror-plane)

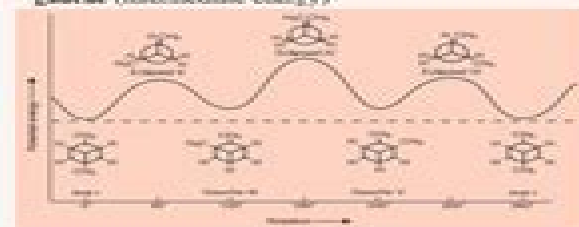


- R/S Notation:** The 4 different atoms or groups attached to a central atom are ranked (a), (b), (c), and (d), by molar mass. The lowest (d) is directed away from the viewer and the sequence of a-b-c produces clockwise (R) or counter-clockwise (S) configuration
- Chiral (Optically Active):** + or - the rotation of plane polarized light, R/S: opposite effects

- Racemic:** 50/50 mixture of stereoisomers (no net optical activity)
- Nomenclature:** Note R/S and +/- in the compound name (e.g., R (+) bromochloromethanol)
- Fisher-Projection:** Diagram depicts chiral/3-D structure
- Molecular Conformations:** Molecule exhibits structural variation due to free rotation about C-C single bond



- Newman Diagram:** Depict rotation about a C-C bond; eclipsed (high energy), anti (low energy), and gauche (intermediate energy)



### Common Terms

**Aliphatic:** Non-aromatic  
**Aromatic:** Benzene ring  
**Conjugation:** Sequence of alternating double (or triple) and single bonds  
**Dielectric Effect:** Polar solvent stabilizes ion formation  
**Exergonic Reaction:** Chemical reaction

that loses energy during the process of the reaction  
**Heterocyclic:** Non-carbon atom in the ring structure  
**Hydrocarbon:** Compound of H and C  
**Ketone:** Molecule made up of a monosaccharide bonded to a ketone

**Monosaccharide:** Carbohydrate that cannot be reduced by hydrolysis into another simple sugar  
**Olefin:** Alkene  
**Paraffin:** Alkane  
**Radioisotope:** Isotope that is radioactive

**Saturated:** Maximum # of Hs (all C-C single bonds)  
**Tetrose:** Monosaccharide with four carbon atoms  
**Unsaturated:** At least one C-C multiple bond  
**Zymology:** The study of the process of fermentation

# Study Guide For Fundamentals Of Organic Chemistry

**T. W. Graham Solomons**



## Study Guide For Fundamentals Of Organic Chemistry:

**Study Guide and Solutions Manual for Fundamentals of Organic Chemistry** Susan McMurry, 1994 Following a brief review of structure and bonding organic molecules and functional groups are presented as early as possible The text is organized primarily by functional group beginning with simple alkanes and moving toward more complex compounds Emphasis is placed on the fundamental mechanistic similarities of organic reactions McMurry's thorough revision continues to present the solid content necessary for this course without sacrifice of important subjects and pedagogical tools Text and reaction summaries full problem sets and outstanding artwork are just some of the features in the Third Edition usually found in a full year book McMurry's clear well written explanations remain a highlight of the book **Organic Chemistry Fundamentals** Speedy Publishing, 2014-07-03 If a student does not have a strong background chemistry and biology or if their knowledge is a little rusty and could use some refreshing a fundamentals guide to organic chemistry is an excellent way to get back into the swing of things A guide will go over the different types of bonds atomic structure and cover the important topics associated with biology that the student will need to understand before entering an organic chemistry class Furthermore study guides are much cheaper than a tutor taking another course or a textbook The guide is succinct too and will not cover unnecessary or ancillary topics that will not be relevant to an organic chemistry course Organic Chemistry Fundamentals Study Guide Speedy Publishing, 2014-11-26 In order to fully understand any subject the fundamentals must be understood and kept in the back of the mind Organic Chemistry is one of the most difficult subjects a college student can take especially if they are not a Chemistry major A lot goes into the fundamentals of the subject That is why an Organic Chemistry Fundamentals book can be so helpful to a student When studying the material if the student discovers they do not understand something they can reference the book and continue with studying in no time at all Having a reference book is the key to success in an Organic Chemistry class Study Guide and Solutions Manual for Fundamentals of Organic Chemistry S. McMurry, 1998-07-07 Fundamentals of Organic Chemistry, 1994 *Organic Chemistry Fundamentals Study Guide* Speedy Publishing LLC, 2014-11-20 In order to fully understand any subject the fundamentals must be understood and kept in the back of the mind Organic Chemistry is one of the most difficult subjects a college student can take especially if they are not a Chemistry major A lot goes into the fundamentals of the subject That is why an Organic Chemistry Fundamentals book can be so helpful to a student When studying the material if the student discovers they do not understand something they can reference the book and continue with studying in no time at all Having a reference book is the key to success in an Organic Chemistry class **Study Guide and Solutions Manual for McMurry's Fundamentals of Organic Chemistry** Speedy Publishing, 2004-03-25 **Organic Chemistry Fundamentals (Speedy Study Guides)** Speedy Publishing LLC, 2014-09-02 If a student does not have a strong background chemistry and biology or if their knowledge is a little rusty and could use some refreshing a fundamentals guide to organic chemistry is an excellent way to get back into the swing of

things A guide will go over the different types of bonds atomic structure and cover the important topics associated with biology that the student will need to understand before entering an organic chemistry class Furthermore study guides are much cheaper than a tutor taking another course or a textbook The guide is succinct too and will not cover unnecessary or ancillary topics that will not be relevant to an organic chemistry course      *Study Guide to Accompany Fundamentals of Organic Chemistry* T. W. Graham Solomons, Jack E. Fernandez, 1986-01-01      000000 0000000000000000 000000, 2017-12

**Fundamentals of Organic Chemistry, Textbook, Study Guide and Solutions Manual** T. W. Graham Solomons, 1996-08-15 A realistic approach to the study of mechanisms The book addresses real functional group chemistry with an emphasis on the biological environmental and medical applications of organic chemistry      **Organic Chemistry Fundamentals (Speedy Study Guide)** Speedy Publishing LLC, 2014-06-30 An Organic Chemistry Fundamentals Reference Guide would be a useful tool for chemistry students medical allied health pharmacy and nursing majors The useful guide has the periodic table of elements varied approaches safety guidelines definitions of terms and common chemical reactions It includes types of organic compounds formulas and isomers nomenclature kinetics and spectroscopy It describes and break downs the basic concepts and applications of organic chemistry in easy to understand language This branch of chemistry deals with the molecular components that make up life on our planet      **Fundamentals of Organic Chemistry** Thomas William Graham Solomons, 1993      **Fundamentals of Organic Chemistry Study Guide and Solutions Manual + Fundamentals of Organic Chemistry** John McMurry, 2006-04-01      *Study Guide with Solutions to Accompany Fundamentals of Organic Chemistry* Ralph J. Fessenden, 1990      **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** McMurry, Susan McMurry, 1990      **Fundamentals of Organic Chemistry** John McMurry, 1986      *Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition* Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines      Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer      *Study Guide and Selected Solutions Manual [for] Fundamentals of General, Organic and Biological Chemistry [by] McMurry ... [et Al.]* Susan McMurry, 2010

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