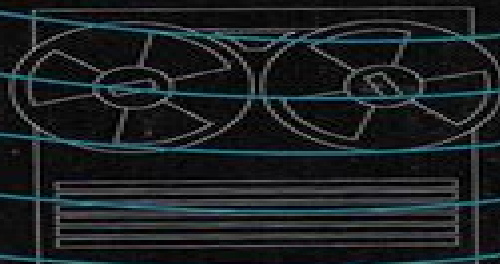
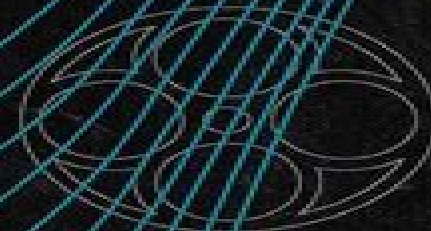
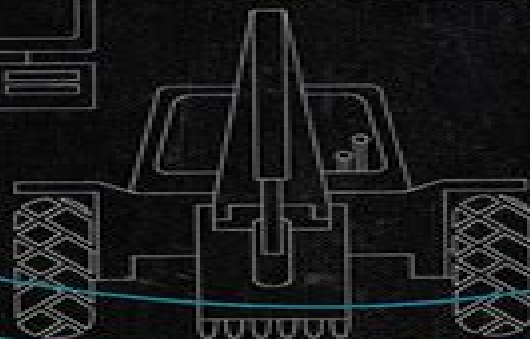


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web jul 12 2023 this concept lies at the heart of all calorimetry problems and calculations because the heat released or absorbed at constant pressure is equal to Δh the relationship between heat and Δh_{rxn} is $\Delta h_{rxn} q_{rxn} q_{calorimeter} m c_s \Delta t$

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web choose 1 answer the final temperature is closer to t_1 than to t_2 a the final temperature is closer to t_1 than to t_2 the final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the final temperature is closer to t_2 than to t_1 c

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