

for GCSE mathematics

# Resource sheets for Intermediate and Higher

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# **Smp Interact Resource And Practice Sheets For N Copymasters**

**Didier Musso** 

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solubility is defined as the maximum quantity of solute that can dissolve in a certain quantity of solvent at a specified temperature differences in the solubility of different solutes are related primarily to differences in the chemical nature of the solutes and the solvents

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1 the solubility of a solute is the maximum mass of the solute that will dissolve in a certain amount of water at a certain this is the same as saying that solubility is the concentration of a unsaturated saturated solution of the solute 2 from your graph what mass of kcl can be dissolved in 100 g of water at these temperatures

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