

Chemical Statics and Kinetics of Solutions

E. A. Moelwyn-Hughes

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The Chemical Statics And Kinetics Of Solutions Physical Chemistry

Yi-Tong Ma



The Chemical Statics And Kinetics Of Solutions Physical Chemistry:

The Chemical Statics and Kinetics of Solutions Emyr Alun Moelwyn-Hughes, 1971 The Journal of Physical Chemistry Wilder Dwight Bancroft, 1910 Includes section New Books **A Text-book of Physical Chemistry, Theory and Practice** Arthur Woolsey Ewell, 1909 *Rydberg Series in Atoms and Molecules* A Duncan, 2012-12-02 Rydberg Series in Atoms and Molecules deals with the Rydberg states of atomic and molecular systems Rydberg series in atomic spectra and of diatomic molecules are examined along with Rydberg terms of polyatomic molecules such as hydrocarbons with triple bonds hydrocarbon molecules with double bonds and benzene and benzene derivatives Comprised of four chapters this book begins with an overview of the spectrum of the hydrogen atom along with some elementary and useful concepts and definitions Some difficulties in the classification of excited states of simple molecules into Rydberg and non Rydberg states are highlighted The next chapter considers Rydberg series in atomic spectra and theoretical methods for quantitative description of atomic structures One electron models for Rydberg terms are also described together with Rydberg series of diatomic molecules such as N₂ and O₂ CO and NO and bromine and iodine The final chapter is devoted to Rydberg terms of polyatomic molecules including hydrocarbons with triple and double bonds benzene and benzene derivatives hydrides with central atom saturated hydrocarbons and aldehydes and ketones This monograph is written primarily for both students and nonspecialists **The Journal of Physical Chemistry**, 1910 **Catalogue** Grove City College, 1921 Electron Densities in Molecular and Molecular Orbitals J.R. Van Wazer, 2012-12-02 Electron Densities in Molecules and Molecular Orbitals aims to explain the subject of molecular orbitals without having to rely much on its mathematical aspect making it more approachable to those who are new to quantum chemistry The book covers topics such as orbitals in quantum chemical calculations electronic ionizations and transitions molecular orbital change distributions orbital transformations and calculations not involving orbitals and electron densities and shapes in atoms and molecules Also included in the book are the cross sectional plots of electron densities of compounds such as organic compounds like methane ethane and ethylene monomeric lithium fluoride and monomeric methyl lithium hydrogen cyanide and methinophosphide and monomeric borane and diborane The text is recommended for those who have begun taking an interest in quantum chemistry but do not wish to deal yet with the mathematics part of the subject **Halogenated Hydrocarbons** A.L. Horvath, 2020-11-25 This book promotes a basic understanding of the concept of solubility and miscibility between halogenated hydrocarbons and water It points out the regularities existing between solubility and physical properties of solute and solvent The book is valuable to chemists and chemical engineers Phases of Matter and their Transitions Gijsbertus de With, 2023-10-30 Phases of Matter and their Transitions An all in one comprehensive take on matter and its phase properties In Phases of Matter and their Transitions accomplished materials scientist Dr Gijsbertus de With delivers an accessible textbook for advanced students in the molecular sciences It offers a balanced and self contained treatment of the thermodynamic and structural aspects of

phases and the transitions between them covering solids liquids gases and their interfaces The book lays the groundwork to describe particles and their interactions from the perspective of classical and quantum mechanics and compares phenomenological and statistical thermodynamics It also examines materials with special properties like glasses liquid crystals and ferroelectrics The author has included an extensive appendix with a guide to the mathematics and theoretical models employed in this resource Readers will also find Thorough introductions to classical and quantum mechanics intermolecular interactions and continuum mechanics Comprehensive explorations of thermodynamics gases liquids and solids Practical discussions of surfaces including their general aspects for solids and liquids Fulsome treatments of discontinuous and continuous transitions including discussions of irreversibility and the return to equilibrium Perfect for advanced students in chemistry and physics Phases of Matter and their Transitions will also earn a place in the libraries of students of materials science

Theory of Unimolecular Reactions Wendell Forst, 2012-12-02 Theory of Unimolecular Reactions provides a comprehensive analysis of the theory of unimolecular reactions also known to kineticists as the Rice Marcus or the Rice Ramsperger Kassel Marcus theory and to those working in mass spectrometry and related fields as the quasi equilibrium theory or the theory of mass spectra This book demonstrates how theoretical parameters are related to experimental observables and describes the methods that are used to obtain useful numerical answers This monograph consists of 11 chapters and begins by explaining the derivation of the expression for the basic rate k_E with emphasis on the unimolecular rate constant intramolecular energy transfer and potential energy surfaces in unimolecular reactions The statistical calculation of unimolecular rate under vibrational potential is also given along with pertinent degrees of freedom The remaining chapters explore the energy distribution functions appropriate to each system the averaging of k_E and the relations between theoretical and experimental parameters Thermal reactions chemical activation systems and the theory of mass spectra are examined The last chapter is devoted to the transition state and its ambiguities This text will be of interest to gas kineticists mass spectrometrists and students and researchers working in the field of physical chemistry

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