Dissolving Salt in Water

VALID ARGUMENT FOR BOTH SIDES

Chemical Change

Dissolving salt may be written as a chemical equation. New products are formed, so a chemical reaction occurs.

 $NaCl(s) \rightarrow Na^{+}(aq) + Cl^{-}(aq)$



Physical Change

Salt doesn't react
with the solvent
(water), so no
chemical reaction
occurs. If you remove
the solvent, the
original compound
remains.

Slow Dissolve

Jason Kalaus



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