

Standard Grade Chemistry
Credit Level
Topic 3
Homework 2

1. (2011C)

A student made the following statements about the particles found in an atom.

A	Relative mass = 1
B	Charge = zero
C	Found outside the nucleus
D	Charge = 1+
E	Charge = 1-

Identify the **two** statements which apply to an electron. (1KU)

2. (2010C)

The grid shows information about some particles.

A $^{23}_{11}\text{Na}$	B $^{18}_8\text{O}$	C $^{40}_{19}\text{K}^+$
D $^{24}_{12}\text{Mg}^{2+}$	E $^{35}_{17}\text{Cl}^-$	F $^{16}_8\text{O}$

(a) Identify the **two** particles with the same number of neutrons. (1KU)

(b) Identify the particle which has the same electron arrangement as neon. (1PS)

3. (2009C)

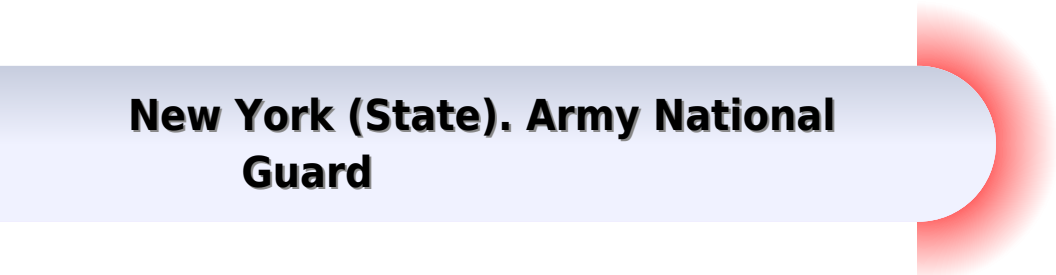
The grid contains information about the particles found in atoms.

A relative mass = 1	B charge = zero	C relative mass almost zero
D charge = 1-	E found outside the nucleus	F charge = 1+

Identify the **two** terms which can be applied to protons. (2KU)

Standard Grade Chemistry With Answers

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Standard Grade Chemistry With Answers:

Standard Grade Chemistry Norman Conquest, Roddy Renfrew, 2002 The second edition of this title for standard grade chemistry matches the requirements of the standard grade and provides coverage of general and credit level material The authors have revised each chapter taking into account of new developments such as in the field of fossil fuels Other chapters have been updated where relevant for instance to reflect changes to the Relative Atomic Mass data The chemical dictionary chapter has been enlarged to include more key words New exam questions have been included together with homework and whole course questions at the end of the book

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Standard Grade Chemistry David Hawley, Sandy McLeman, 2012-07-25

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About the author s David is an experienced PT and Chemistry author He is also the SQA s Principal Examiner at Higher

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