

OVER 500 SONGS INCLUDING THESE GREAT STANDARDS...

ALFIE
DANCE IN WONDERLAND
APRIL IN PARIS
DUMIN IN NEW YORK
E' CAME MUCHO
BLACK COFFEE
BRAZIL
CARAVAN
OUR FATE TO THE WIND
LAWDRY 'BOUT ME
THE FLINTSTONES
MAGIA ON MY MIND
GIRL TALK
GRAND WALTZ
DEEP IS THE OCEAN
DIED ON THE MOON

I'VE GOT THE WORLD ON A STRING
IN A SENTIMENTAL MOOD
IN THE WEE SMALL HOURS OF THE
MORNING
ISN'T IT ROMANTIC?
JITTERBUG WALTZ
LOVER
MAISON' WHOOREE!
MISSION: IMPOSSIBLE THEME
MOOD INDIGO
MOON RIVER
MY HEART STOOD STILL
MY OLD FLAME
NATURE BOY
NIGHT TRAIN
NORWEGIAN WOOD

OUT OF NOWHERE
THE RAINBOW CONNECTION
THE SHADOW OF YOUR SMILE
SOFTLY AS IN A MORNING SUNRISE
SOMEBODY LOVES ME
SOPHISTICATED LADY
STAR DUST
STELLA BY STARLIGHT
TAKE FIVE
TANGERINE
THIS MASQUERADE
TOO LATE NOW
THE VERY THOUGHT OF YOU
WATERMELON MAN
WAVE
WHEN SUNNY GETS BLUE

...AND THESE JAZZ CLASSICS!

500 MILES HIGH / CHICK COREA
A LA DE BERER / ANTONIO CARLOS JOBIM
CROLOGY / CHARLIE PARKER, DIZZY GILLESPIE
SAX FOR BARKSDALE / GERRY MULLIGAN
BLUEPORT / ART FARMER
BLUES FOR ALICE / CHARLIE PARKER
EYES FOR JUNIOR (PYRAMID) / RAY BROWN
BLUES IN TIME / PAUL DESMOND
CANNONBALL / JOSEF ZAWINUL
CAREFUL / JIM HALL
CHAMELEON / HERBIE HANCOCK
CHANGE OF SEASON / JOHN RATTUCCI
CHROMOZONE / MIKE STERN
GO AHEAD / RUSS FREEMAN, THE RIPPINGTONS
DAT DERE / BOBBY TIMMONS
FOUR AHEAD / HERB ELLIS, JOHNNY FRIGO
DOLPHIN DANCE / HERBIE HANCOCK
EPISTROPHY / THELONIOUS MONK
FEELS SO GOOD / CHUCK MANGIONE
FILTHY McNASTY / HORACE SILVER
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FREEDOM JAZZ DANCE / EDDIE HARRIS
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I MEAN YOU / THELONIOUS MONK

IF WE MEET AGAIN, PART ONE / AL DI MEOLA
IMPRESSIONS / JOHN COLTRANE
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LITTLE SUNFLOWER / FREDDIE HUBBARD
MAIDEN VOYAGE / HERBIE HANCOCK
MANTECA / DIZZY GILLESPIE
MAS QUE NADA / BRAZIL '66
MOMENT'S NOTICE / JOHN COLTRANE
MORNING DANCE / SPYRO GYRA
MY ATTORNEY BERNIE / DAVE FRISHBERG
NUAGES / DJANGO REINHARDT
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Table of Contents The Hal Leonard Little Real Jazz C Edition

1. Understanding the eBook The Hal Leonard Little Real Jazz C Edition
 - The Rise of Digital Reading The Hal Leonard Little Real Jazz C Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying The Hal Leonard Little Real Jazz C Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an The Hal Leonard Little Real Jazz C Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from The Hal Leonard Little Real Jazz C Edition

- Personalized Recommendations
- The Hal Leonard Little Real Jazz C Edition User Reviews and Ratings
- The Hal Leonard Little Real Jazz C Edition and Bestseller Lists
- 5. Accessing The Hal Leonard Little Real Jazz C Edition Free and Paid eBooks
 - The Hal Leonard Little Real Jazz C Edition Public Domain eBooks
 - The Hal Leonard Little Real Jazz C Edition eBook Subscription Services
 - The Hal Leonard Little Real Jazz C Edition Budget-Friendly Options
- 6. Navigating The Hal Leonard Little Real Jazz C Edition eBook Formats
 - ePub, PDF, MOBI, and More
 - The Hal Leonard Little Real Jazz C Edition Compatibility with Devices
 - The Hal Leonard Little Real Jazz C Edition Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of The Hal Leonard Little Real Jazz C Edition
 - Highlighting and Note-Taking The Hal Leonard Little Real Jazz C Edition
 - Interactive Elements The Hal Leonard Little Real Jazz C Edition
- 8. Staying Engaged with The Hal Leonard Little Real Jazz C Edition
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers The Hal Leonard Little Real Jazz C Edition
- 9. Balancing eBooks and Physical Books The Hal Leonard Little Real Jazz C Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection The Hal Leonard Little Real Jazz C Edition
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine The Hal Leonard Little Real Jazz C Edition
 - Setting Reading Goals The Hal Leonard Little Real Jazz C Edition
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of The Hal Leonard Little Real Jazz C Edition

-
- Fact-Checking eBook Content of The Hal Leonard Little Real Jazz C Edition
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely becuase bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno₃ h₂so₄ in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no₂⁺ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of

nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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