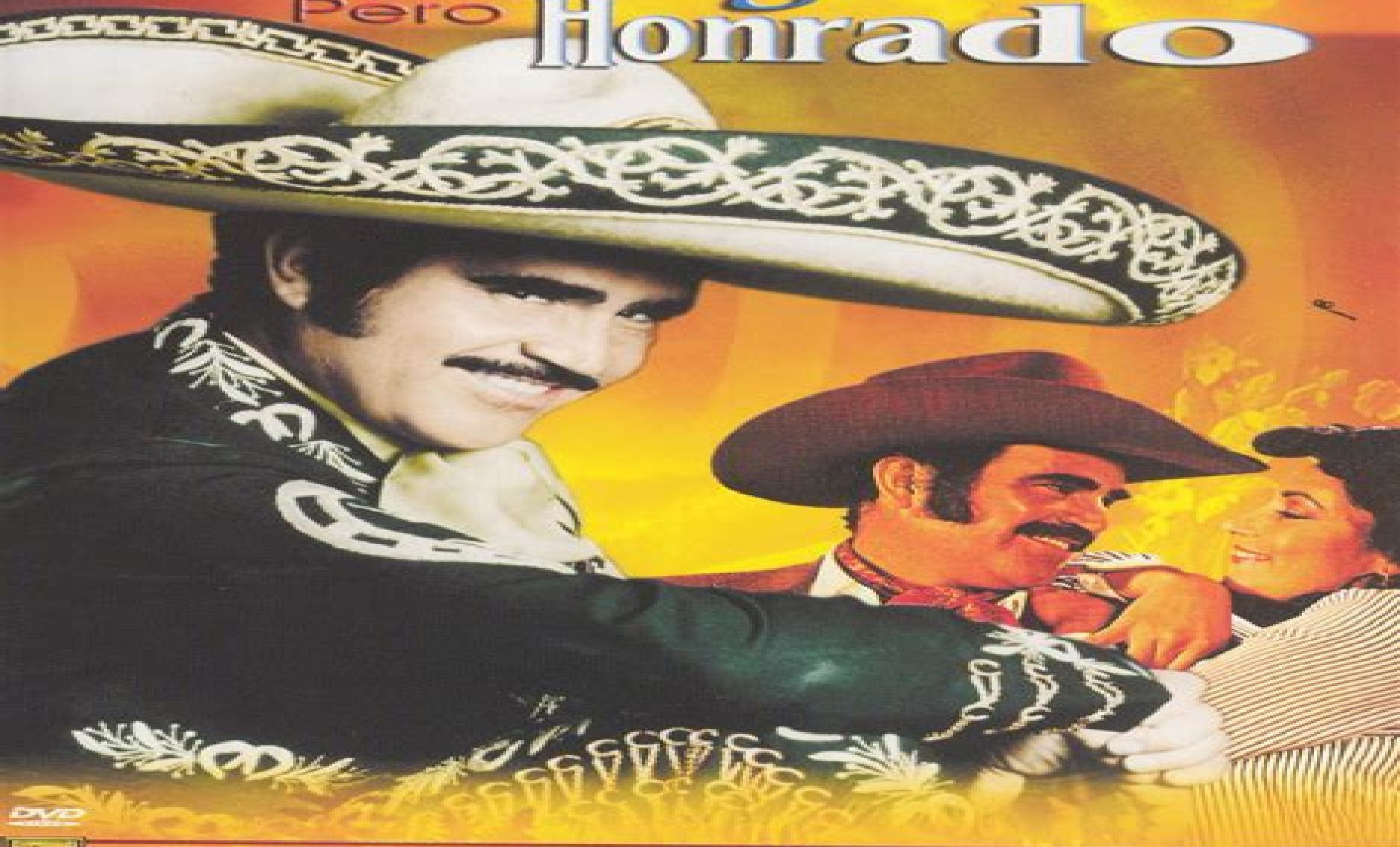


VICENTE FERNANDEZ

# *Sinvergüenza* Pero Honrado



DVD



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# Sinverguenza Pero Honrado

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will be obtained from a 0.1138 g sample of alloy which is 73.2% Al. At  $P = 754.3 \text{ torr}$  and  $T = 21.2^\circ\text{C}$ , what volume in mL will the dry hydrogen gas from problem 1 occupy at STP?

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web we will exploit the fact that the amount of hydrogen produced by one gram of zinc is different from the amount of hydrogen produced by one gram of aluminum. 1 mol zinc 65.4 g 1 mol of hydrogen gas 1.00 g Zn 0.0153 mol  $\text{H}_2$  3.2 mol aluminum 54.0 g 3 mol hydrogen gas 1.00 g Al 0.0556 mol  $\text{H}_2$  4. we can react an alloy of

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