

BRIAN W. PFENNIG

# PRINCIPLES OF INORGANIC CHEMISTRY

SECOND EDITION



WILEY

# The Principles Of Inorganic Chemistry

**Robert B. Jordan**



## **The Principles Of Inorganic Chemistry:**

**The Principles of Inorganic Chemistry** William L. Jolly, 1976      *Principles of Inorganic Chemistry* Robert B. Jordan, 2024-04-22 This textbook provides a current and comprehensive coverage of all major topics of inorganic chemistry in a single source. It includes an analysis of the sources and preparations of the elements, their common compounds, their aqueous speciation, and their applications, while it also discusses reaction pathways and mechanisms. It includes up-to-date material supported by over 4000 references to the original literature and to recent reviews that provide more detailed information. The material is accompanied by over 250 figures and three-dimensional representations based on published structural details. Each chapter has worked examples and problems with multiple inserts describing topical issues related to the material in the text. The textbook provides the instructor with a wide range of areas that can be selected to meet the background and interests of the students, while selected chapters are relevant to courses on more specialized topics such as inorganic materials, bioinorganic chemistry, and nanomaterials. The intended readers are students, lecturers, and researchers who need a source for the current status of the area.      PRINCIPLES OF INORGANIC CHEMISTRY WILHELM. OSTWALD, 2018      **Principles of Inorganic Chemistry** Brian W. Pfennig, 2015-03-03 Aimed at senior undergraduates and first-year graduate students, this book offers a principles-based approach to inorganic chemistry that, unlike other texts, uses chemical applications of group theory and molecular orbital theory throughout as an underlying framework. This highly physical approach allows students to derive the greatest benefit of topics such as molecular orbital acid-base theory, band theory of solids, and inorganic photochemistry, to name a few. Takes a principles-based group and molecular orbital theory approach to inorganic chemistry. The first inorganic chemistry textbook to provide a thorough treatment of group theory, a topic usually relegated to only one or two chapters of texts, giving it only a cursory overview. Covers atomic and molecular term symbols, symmetry, coordinates in vibrational spectroscopy using the projection operator method, polyatomic MO theory, band theory, and Tanabe-Sugano diagrams. Includes a heavy dose of group theory in the primary inorganic textbook. Most of the pedagogical benefits of integration and reinforcement of this material in the treatment of other topics such as frontier MO, acid-base theory, band theory of solids, inorganic photochemistry, the Jahn-Teller effect, and Wade's rules are fully realized. Very physical in nature, compare to other textbooks in the field, taking the time to go through mathematical derivations and to compare and contrast different theories of bonding in order to allow for a more rigorous treatment of their application to molecular structure, bonding, and spectroscopy. Informal and engaging writing style, worked examples throughout the text, unanswered problems in every chapter, contains a generous use of informative, colorful illustrations.      The Principles of Inorganic Chemistry Wilhelm Ostwald, 1902      **Basic Principles of Inorganic Chemistry** Brian J. Hathaway, Clair Murphy, Brian Murphy, 2007-10-31 General chemistry textbooks are usually lengthy and present chemistry to the student as an unconnected list of facts. In inorganic chemistry, emphasis should be placed on the connections between valence shell

electron configuration and the physical and chemical properties of the element Basic Principles of Inorganic Chemistry Making the Connections is a short concise book that emphasises these connections in particular the chemistry of the Main Group compounds With reference to chemical properties Lewis Structures stoichiometry and spider diagrams students will be able to predict or calculate the chemistry of simple polyatomic compounds from the valence shell configuration and will no longer be required to memorise vast amounts of factual chemistry This book is ideal for students taking chemistry as a subsidiary subject as well as honours degree students **The Principles of Inorganic Chemistry** Wilhelm

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