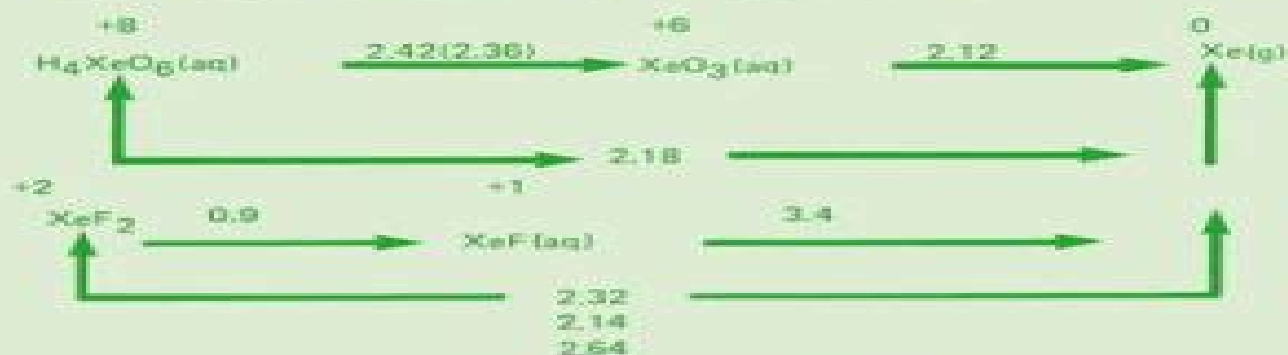


STANDARD POTENTIALS IN AQUEOUS SOLUTION



edited by Allen J. Bard
Roger Parsons
Joseph Jordan



Prepared under the auspices of IUPAC

Standard Potentials In Aqueous Solution

Gary Wulfsberg



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The present Supplement Volume Beryllium A 3 continues and completes the description of the physical properties of the element begun in Supplement Volume A 2 1991 and also treats the electrochemical behavior of the metal The unique combination of the Be properties which was pointed out in Supplement Volume A 2 is also demonstrated in the following chapters of this Volume A 3 13 Electrical Properties 14 Electronic Properties 15 Optical Properties Emission and Impact Phenomena 16 Electrochemical Behavior Starting with the electrical properties Be is a rather good electrical conductor in contrast to what might be expected Superconductivity was studied especially on films Quantum effects which are more pronounced in Be than in most other metals are the reason for numerous investigations of the magnetoresistance and the magnetic breakdown effect The basis for many of the characteristic properties is the unique nature of bonding in Be as a consequence of its peculiar electronic structure and the special shape of its Fermi surface which also gave rise to further numerous studies Detailed cluster calculations were performed to better understand the bonding in the metal Regarding the optical properties the high reflectivity of Be particularly in the infrared region makes it attractive for the fabrication of precision optical surfaces mirrors it is also useful for solar collector surfaces in spacecraft applications Emission and electron and ion impact phenomena as well as neutron optics are also discussed Inorganic Chemistry Gary Wulfsberg, 2000-03-16 Both elementary inorganic reaction chemistry and more advanced inorganic theories are presented in this one textbook while showing the relationships between the two **Quanta, Matter, and Change** Peter Atkins, Julio de Paula, Ronald Friedman, 2009 aspects of the learning process are fully supported including the understanding of terminology notation mathematical concepts and the application of physical chemistry to other branches of science Building on the heritage of the world renowned Atkins Physical Chemistry *Quanta Matter and Change* gives a refreshing new insight into the familiar by

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