

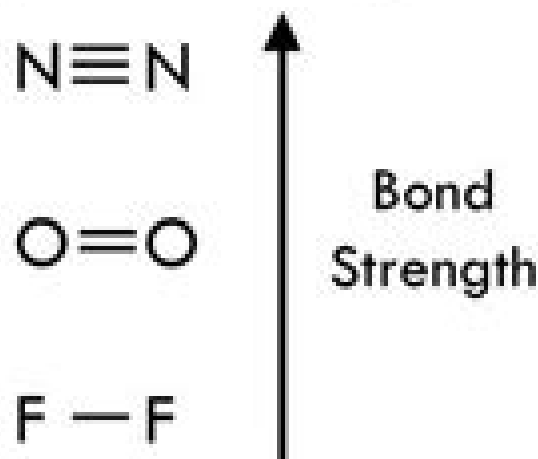
Bond Strengths

50

TABLE 2.2 Bond Dissociation Energies of Diatomic Molecules ($\text{kJ}\cdot\text{mol}^{-1}$)

Molecule	Bond dissociation energy
H_2	424
N_2	932
O_2	484
CO	1062
F_2	146
Cl_2	230
Br_2	181
I_2	139
HF	543
HCl	419
HBr	354
HI	287

- Diatomic Molecules
- Trends can be explained from Lewis structures
- Higher bond orders are stronger
- A Triple bond tends to be stronger than a single bond



Strengths Of Chemical Bonds 2nd Edition

D Kirk

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