## Electronic Theory of Valency

A chemical bonding is the force which hold the atoms of molecule together. Bond formation is exothermic process i.e. decrease in energy of system.

- √ Occurs when force of attraction is greater than force of repulsion.
- The no. of electron lost, gained or shared by atom of an element is known as valency.

#### Electrovalent or Ionic Bond

- A bond formed by the transference of electron from atom to another such that both combining atom attain similar stable configuration of their nearest inert element.
- The number of electron involved (i.e. transferred) is called electro-valency of atom and compound thus formed is known as electrovalent compound.
- In the formation of electrovalent bond one of the atoms loses electron and acts as electropositive, element while other atom gain electron and act as electroneostive element.
- Generally, the electropositive (cation) element are metal are carry positive charge and electronegative element are non- metal and carry negative charge (anion)
- √ The oppositely charged ions are held together by electrostatic force of attraction.
- V lonic bond is favoured by

low ionization energy of cation high electron affinity of atoms forming anion high lattice of energy of compound

#### Properties of Ionic Compound

- √ They exist in solid
- √ They are soluble in polar solvent like water and insoluble non-polar solvent like ether, benzene etc.
- Since, ions in a molecule held together by a strong force of attraction. So they have high m.pt and b.pt.
- Due to presence of free mobile ions, ionic compound conduct heat and electricity in aqueous and molten state.
- V NaCl do not conduct heat because their ions held together by strong force of attraction.
- √ Ionic compound are non-directional in nature and don't stereoisomerism.
- √ Ionic compounds are ionic and not molecular.

#### Covalent bond

- √ Formed by mutual sharing of electrons between the combing atoms.
- The number of valence electron involved in the formation of covalent bond is termed as covalency and the compound thus formed is known as covalent compound.

# **The Electronic Theory Of Valency**

**Frank Johnson Welcher** 

 $\ \, \textbf{The Electronic Theory Of Valency:} \\$ 

Whispering the Strategies of Language: An Emotional Quest through The Electronic Theory Of Valency

In a digitally-driven world wherever displays reign great and immediate communication drowns out the subtleties of language, the profound secrets and emotional subtleties concealed within words usually get unheard. However, nestled within the pages of **The Electronic Theory Of Valency** a captivating fictional prize sporting with fresh emotions, lies a fantastic quest waiting to be undertaken. Penned by an experienced wordsmith, this wonderful opus invites readers on an introspective journey, delicately unraveling the veiled truths and profound affect resonating within the very cloth of every word. Within the mental depths with this touching review, we will embark upon a genuine exploration of the book is core subjects, dissect their charming publishing fashion, and yield to the effective resonance it evokes strong within the recesses of readers hearts.

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