

El sistema periódico

Gracias al descubrimiento de la electrolisis, se descubrieron una serie de elementos químicos. Gracias a la espectroscopia, hubo un gran avance en el descubrimiento de los elementos químicos. Actualmente se conocen 118 elementos. Una vez empezó el descubrimiento de los elementos químicos, se empezó a ver cómo podemos clasificarlos. La primera clasificación la debemos a **Lavoisier**. Ordenó los 33 "elementos" conocidos (además de la luz, el calórico y algunos compuestos) en una tabla de sustancias simples, que fue el primer esbozo de una tabla ordenada de los elementos químicos.

Un elemento químico es toda sustancia que los químicos no puedan separar en otras más simples por medios químicos.

GRUPOS	ELEMENTOS
Sin grupo	Luz
Grupo 1: Sustancias que pueden ser los elementos de los cuerpos	Calórico, oxígeno, azoe, hidrógeno
Grupo 2: Sustancias no metálicas, oxidables y acidificables	azufre, fósforo, carbono, radical muriático, radical fluorico, radical borácico
Grupo 3: Sustancias metálicas, oxidables y acidificables	antimonio, plata, arsénico, bismuto, cobalto, cobre, estaño, hierro, manganeso, mercurio, molibdeno, nickel, oro, platino, plomo, tungsteno, zinc
Grupo 4: Salificables térreas	cal, magnesia, benta, alúmina, sílica

Poco después **Dobereiner** escribió la **ley de las triadas**. Manifestó la existencia de grupos de tres elementos, triadas, que presentaban analogías en sus propiedades físicas y químicas. Ordenadas por sus masas atómicas, la masa del elemento central era aproximadamente la media aritmética de las masas de los otros dos.

ELEMENTOS	MASAS ATÓMICAS	MEDIA	PROPIEDADES
Li Na K	7 23 39	23	Metales muy reactivos
Ca Sr Ba	40 88 137	88,5	Metales reactivos
S Se Te	32 79 126	79	No metales coloreados
Cl Br I	35 80 127	81	No metales muy reactivos

Congreso de Karlsruhe

Celebrado en 1860. Fue el primer congreso internacional de Química. Se estableció claramente la diferencia entre los conceptos de átomo y molécula.

Cannizzaro defendió la hipótesis de Avogadro, hasta entonces desconocida por la mayoría de los presentes. En base a dicha hipótesis, presentó una nueva tabla de pesos atómicos, que sirvió de base para el rápido desarrollo posterior de la tabla periódica.

Sistema Periodico El

Ying-Ying Zheng



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Table of Contents Sistema Periodico El

1. Understanding the eBook Sistema Periodico El
 - The Rise of Digital Reading Sistema Periodico El
 - Advantages of eBooks Over Traditional Books
2. Identifying Sistema Periodico El
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sistema Periodico El
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sistema Periodico El
 - Personalized Recommendations
 - Sistema Periodico El User Reviews and Ratings
 - Sistema Periodico El and Bestseller Lists
5. Accessing Sistema Periodico El Free and Paid eBooks
 - Sistema Periodico El Public Domain eBooks
 - Sistema Periodico El eBook Subscription Services
 - Sistema Periodico El Budget-Friendly Options

6. Navigating Sistema Periodico El eBook Formats
 - ePub, PDF, MOBI, and More
 - Sistema Periodico El Compatibility with Devices
 - Sistema Periodico El Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sistema Periodico El
 - Highlighting and Note-Taking Sistema Periodico El
 - Interactive Elements Sistema Periodico El
8. Staying Engaged with Sistema Periodico El
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sistema Periodico El
9. Balancing eBooks and Physical Books Sistema Periodico El
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sistema Periodico El
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sistema Periodico El
 - Setting Reading Goals Sistema Periodico El
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sistema Periodico El
 - Fact-Checking eBook Content of Sistema Periodico El
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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