

The Periodic Table of the Elements

The Periodic Table of the Elements

The periodic table displays elements organized by atomic number (1 to 118) and grouped into categories. The callout for Iron (Fe) provides the following data:

- atomic mass: 55.845
- atomic number: 26
- electronegativity: 1.83
- chemical symbol: Fe
- name: Iron
- electron configuration: [Ar] 3d⁶ 4s²
- oxidation states: most common are +2

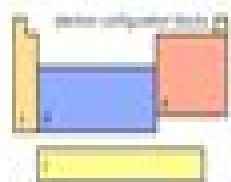
Legend:

- alkali metals
- alkaline earth metals
- other metals
- transition metals
- lanthanoids
- actinoids
- metalloids
- nonmetals
- halogens
- noble gases
- unknown elements
- radioactive elements (indicated by a yellow symbol)

atomic mass: 55.845
 1st ionization energy: 7.62 eV
 electronegativity: 1.83
 chemical symbol: Fe
 name: Iron
 electron configuration: [Ar] 3d⁶ 4s²

alkali metals
 alkaline metals
 other metals
 transition metals
 lanthanoids
 actinoids

metalloids
 nonmetals
 halogens
 noble gases
 unknown elements
 radioactive elements (not shown in periodic table)



Notes:
 1. Not all elements 113-118 have an official name designation (see table).
 2. 113-118 are in the 7th period.
 3. All elements are subject to future revision (see table).

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No

The Periodic Table

Judd E. Hollander



The Periodic Table:

The Periodic Table Eric R. Scerri, 2020 The Periodic Table Its Story and Its Significance traces the evolution and development of the periodic table from Mendeleev's 1869 first published table and onto the modern understanding provided by modern physics

Mystery of the Periodic Table Benjamin D Wiker, 2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today He introduces the young reader to people like Von Helmholtz, Boyle, Stahl, Priestley, Cavendish, Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day The first part of Wiker's witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages 12-13 and up with a final chapter somewhat more advanced Illustrated by Jeanne Bendick and Ted Schluenderfritz

The Periodic Table Tom Jackson, 2020-02-04 Which is the densest element Which has the largest atoms And why are some elements radioactive From the little known uses of gold in medicine to the development of the hydrogen bomb this is a fresh new look at the Periodic Table Combining cutting edge science with fascinating facts and stunning infographics this book looks at the extraordinary stories of discovery amazing properties and surprising uses of each element whether solid liquid or gas naturally occurring synthesised or theoretical From hydrogen to Oganesson this is a fact filled visual guide to each element each accompanied by technical data category atomic number weight boiling point as well as fun facts and stories about their discovery and surprising uses

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Leveled Texts: The Periodic Table Joshua Bishop-Roby, 2014-01-01 All students can learn about the periodic table through text written at four different reading levels Symbols on the pages represent reading level ranges to help differentiate instruction Provided comprehension questions complement the text

The Periodic Table Paul Parsons, 2014-03-11 As one of the most recognizable images in science the periodic table is ingrained in our culture First drawn up in 1869 by Dmitri Mendeleev its 118 elements make up not only everything on our planet but also everything in the entire universe The Periodic Table looks at the fascinating story and surprising uses of each of those elements whether solid liquid or gas From the little known uses of gold in medicine to the development of the hydrogen bomb each entry is accompanied by technical data category atomic number weight boiling point presented in easy to read headers and a colour coding system that helps the reader to navigate through the different groups of elements A remarkable display of thought provoking science and beautiful photography this guide will allow the reader to discover the world afresh

How the

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