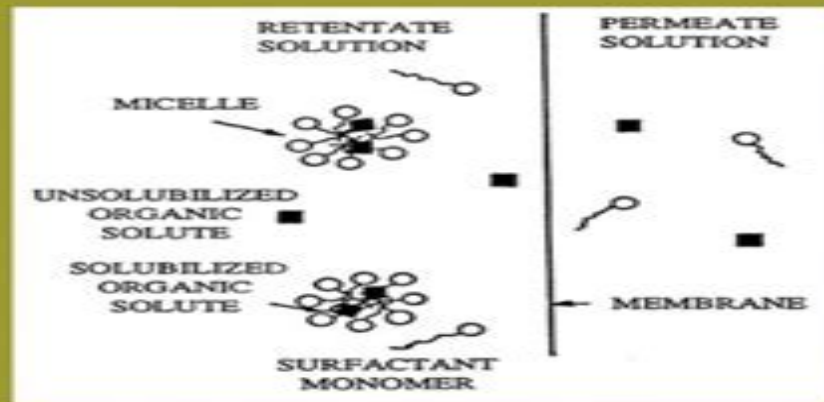


surfactant science series
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SOLUBILIZATION IN SURFACTANT AGGREGATES



edited by
Sherril D. Christian
John F. Scamehorn



CRC Press
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Solubilization In Surfactant Aggregates

P. Somasundaran



Solubilization In Surfactant Aggregates:

Solubilization in Surfactant Aggregates Sherril D. Christian, John F. Scamehorn, 1995-04-10 This work covers topics ranging from fundamental studies of solubilization to practical technological applications of the phenomenon It reviews the solubilization of organic materials into surfactant aggregates including micelles vesicles and admicelles The book also details methods of measuring solubilization that utilize both classical and newer instrumental techniques It is intended for physical surface colloid and surfactant chemists chemical environmental and civil engineers and upper level undergraduate and graduate students in these disciplines Adsorption and Aggregation of Surfactants in Solution K.L. Mittal, Dinesh O.

Shah, 2002-11-07 Offering the latest research and developments in the understanding of surfactant behavior in solutions this reference investigates the role and dynamics of surfactants and their solution properties in the formulation of paints printing inks paper coatings pharmaceuticals personal care products cosmetics liquid detergents and lubricants Exploring the science behind techniques from oil recovery to drug delivery the book covers surfactant stabilized particles solid particles at liquid interfaces nanocapsules aggregation behavior of surfactants micellar catalysis vesicles and liposomes the clouding phenomena viscoelasticity of micellar solutions and more Design and Applications of Self-Assembly Aggregates - From Micelles to Nanoemulsions Juan Mejuto, Mihalj Poša, 2024-10-23 Self assembled colloidal aggregates are made up of nano or micrometer sized particles dispersed in a continuous phase that organize into ordered structures due to intrinsic physical and chemical interactions like electrostatic forces hydrophobic hydrophilic interactions Van der Waals forces and hydrogen bonds These systems are stable and form a wide variety of structures including micelles vesicles liquid crystals and emulsions Their ability to create sophisticated materials makes them valuable in various fields including materials science pharmacology biotechnology medicine food technology and cosmetics Despite their advantages challenges remain in achieving precise control over the self assembly process Design and Applications of Self Assembly Aggregates From Micelles to Nanoemulsions is a collaborative effort by different authors exploring research on these microheterogeneous systems and their diverse applications *Equations of State for Fluids and Fluid Mixtures* J.V. Sengers, R.F. Kayser, C.J. Peters, H.J. White, 2000-10-30

This book has been prepared under the auspices of Commission I 2 on Thermodynamics of the International Union of Pure and Applied Chemistry IUPAC The authors of the 18 chapters are all recognized experts in the field The book gives an up to date presentation of equations of state for fluids and fluid mixtures All principal approaches for developing equations of state are covered The theoretical basis and practical use of each type of equation is discussed and the strength and weaknesses of each is addressed Topics addressed include the virial equation of state cubic equations and generalized van der Waals equations perturbation theory integral equations corresponding states and mixing rules Special attention is also devoted to associating fluids polydisperse fluids polymer systems self assembled systems ionic fluids and fluids near critical points

Handbook of Surface and Colloid Chemistry K. S. Birdi, 2008-11-20 The third edition of this bestseller covers the latest

advancements in this rapidly growing field Focusing on analyses and critical evaluation of the subject this new edition reviews the most up to date research available in the current literature International contributors offer their perspectives on various topics including micellar systems mi

Surfactant Formulation Engineering using HLD and NAC Edgar Acosta, Jeffrey Harwell, David A. Sabatini, 2025-04-28 Surfactants are molecules that contain groups that are water loving hydrophilic and oil loving lipophilic The central question in formulations is often which of the two portions dominate the behavior of the surfactant For many years that question was answered in terms of the surfactant structure only However the modern view is that the hydrophilic lipophilic nature of the surfactant is the result of surfactant structure and formulation conditions nature of the oil temperature aqueous phase composition as captured by a semi empirical equation called the hydrophilic lipophilic difference HLD The HLD is a dimensionless number that indicate the approach to the point where the surfactant inverts its solubility from being water soluble negative HLD to oil soluble positive HLD The HLD alone is a good indicator of how the formulation could behave but it does not produce any formulation property that can be used to predict product performance The net average curvature NAC are a set of equations that take the value of HLD to predict the properties of the formulation such as oil and or water solubilization capacity interfacial tension phase diagrams contact angle and others Surfactant Formulation Engineering using HLD and NAC will not only introduce the reader to HLD NAC but also to the practical use of these concepts in numerous applications ranging from application in the petroleum industry to environmental remediation to food cosmetic and pharmaceutical applications and even nanotechnology The last part of the book will look at the molecular origins of the empirical terms in HLD via the Integrated Free Energy Model IFEM Concentrates on the HLD and NAC providing industrially relevant examples Provides the only single depository for HLD parameters Balances theory and application with insights from both academic and industrial authors Includes examples relevant to a wide range of fields with practical guides on how to go from the formulation objective s to an actual formulation design

Surfactants Bob Aveyard, 2019-10-09 Surfactants today you have probably eaten some or rubbed others on your body Plants animals including you and microorganisms make them and many everyday products e g detergents cosmetics foodstuffs contain them Surfactant molecules have one part which is soluble in water and another which is not This gives surfactant molecules two valuable properties 1 they adsorb at surfaces e g of an oil droplet in water and 2 they stick together aggregate in water The aggregates micelles are able to dissolve materials not soluble in water alone and adsorbed surfactant layers at the surfaces of particles or say oil droplets in water stop the particles or drops sticking together This is why stable emulsions such as milk do not separate into layers This book treats the basic physical chemistry and physics underlying the behaviour of surfactant systems In this book you will first learn about some background material including hydrophobic hydration interfacial tension and capillarity Section I Discussion of surfactant adsorption at liquid fluid and solid liquid interfaces is given in Section II and includes thermodynamics of adsorption dynamic and rheological aspects of liquid

interfaces and the direct characterisation of surfactant monolayers In Section III a description is given of surfactant aggregation to give micelles lyotropic liquid crystals microemulsions and Winsor systems There follows a discussion of surface forces and the way they confer stability on lyophobic colloids and thin liquid films Section IV Various dispersions stabilised by adsorbed surfactant or polymer including solid in liquid dispersions emulsions and foams are considered in Section V The wetting of solids and liquids is explored in Section VI Like surfactants small solid particles can adsorb at liquid fluid interfaces form monolayers and stabilise emulsions and foams Such behaviour is covered in Section VII It is assumed the reader has a knowledge of undergraduate physical chemistry particularly chemical thermodynamics and of simple physics Mathematics elementary algebra and calculus is kept at a level consistent with the straightforward derivation of many of the equations presented **Encyclopedia of Surface and Colloid Science** P. Somasundaran,2006

Structure-Performance Relationships in Surfactants Kunio Esumi,Minoru Ueno,2003-03-18 In response to intensifying interest on surfactant research brought on by recent innovation Structure Performance Relationships in Surfactants Second Edition examines novel developments in our understanding of the properties and performance of surfactants at air liquid liquid liquid and solid liquid interfaces highlighting seven new chapters and carefully updated material to reflect current trends This edition presents new material on the adsorption of vesicle forming surfactants at the air water interface fluorinated surfactants having two hydrophobic chains surface active properties of telomer type surfactants having several hydrocarbon chains and the association behavior of amphiphilic dendritic polymers among many other topics *Mixed Surfactant Systems* Masahiko Abe,2004-12-22 Completely revised and expanded throughout Mixed Surfactant Systems Second Edition surveys the latest results newest experimental perspectives and theoretical investigations of properties behavior and techniques applicable to mixed surfactant systems This important book elucidates core theoretical notions while summarizing results of

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