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CHARACTER VARIETY OF REPRESENTATIONS OF A FINITELY GENERATED GROUP IN SL_2

KYOJI SAITO

(Received January 10, 1996)

This is a partial exposition of [S4, S5, S6], which study the space of representations of a (finitely generated) group Γ in SL₂ and GL₂ in an attempt of its application in geometry: Teichmüller spaces, knot theory, hyperbolic manifolds, moduli spaces, · · · , etc. (see for instance, [A], [Be], [Br], [C-C-G-L-S], [C-S], [F-K], [G], [H], [H-L-M], [H-K], [J-W], [K], [Kj], [Ko], [Kr], [Ma], [Mu], [N-Z], [O], [S], [S-S], [T], [W], [We], [Wo], [Y], · · · , etc). For simplicity, we omit the case for GL₂ in the present exposition. Let us explain the main result of the present paper.

Let Γ be a group. The purpose of the present paper is to introduce the character variety $Ch(\Gamma, SL_2)$ in order to parameterize conjugacy classes of representations of Γ in SL_2 in a functorial way. At first, the character variety is introduced as a scheme over \mathbb{Z} , independent of the coefficient ring of representations in question. Then, the scaler field is specialized to \mathbb{R} to obtain results on representations in $SL_2(\mathbb{R})$ and in SU(2) with respect to the classical topology. Let us explain this briefly.

Let $\operatorname{Hom}(\Gamma,\operatorname{SL}_n)$ be the functor $R \in \{\operatorname{commutative rings with } 1\} \mapsto \operatorname{Hom}(\Gamma,\operatorname{SL}_n(R)) \in \{\operatorname{sets}\}$. The functor is representable (see §1.3 Lemma) and so, for an abuse of notation, we denote by the same $\operatorname{Hom}(\Gamma,\operatorname{SL}_n)$ the scheme over $\mathbb Z$ representing the functor. The group scheme PGL_n acts on $\operatorname{Hom}(\Gamma,\operatorname{SL}_n)$. Whether the universal categorical quotient $\operatorname{Hom}(\Gamma,\operatorname{SL}_n)//\operatorname{PGL}_n$ (Mumford [Mu1]) defined over $\mathbb Z$ exists or not seems to be a hard and unsolved question. Instead of asking directly for the quotient space, we introduce i) the character variety $\operatorname{Ch}(\Gamma,\operatorname{SL}_2)$ together with its discriminant subvariety D_Γ as schemes over $\mathbb Z$ abstractly, and ii) the PGL_2 -invariant morphism π_Γ : $\operatorname{Hom}(\Gamma,\operatorname{SL}_2) \to \operatorname{Ch}(\Gamma,\operatorname{SL}_2)$, for which we prove that i) the restriction of π_Γ on the complement of $\pi_\Gamma^{-1}(D_\Gamma)$ is a principal PGL_2 -bundle with respect to the etal topology, and ii) the inverse image $\pi_\Gamma^{-1}(D_\Gamma)$ is a subfunctor of $\operatorname{Hom}(\Gamma,\operatorname{SL}_2)$ consisting of abelian or reducible representations. This implies that the complement $\operatorname{Hom}^*(\Gamma,\operatorname{SL}_2) := \operatorname{Hom}(\Gamma,\operatorname{SL}_2) \setminus \pi^{-1}(D_\Gamma)$ consists of absolutely irreducible representations, and that $\operatorname{Hom}^*(\Gamma,\operatorname{SL}_2)$ has the universal categorical quotient space $\operatorname{Ch}^*(\Gamma,\operatorname{SL}_2) := \operatorname{Ch}(\Gamma,\operatorname{SL}_2) \setminus D_\Gamma$ defined over $\mathbb Z$. Then the result is specialized

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2 o 3 4al 3o 2 2al 2 o 3 4 n 2 h 2 nh 3 n

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web chemical equations and the law of conservation of matter in the previous section the reaction between hydrogen gas and oxygen gas to produce water in the gaseous phase was shown as a chemical equation h 2 g o 2 g h 2 o g at the molecular level the reaction would look something like this notice that there are two oxygen atoms on the

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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined the first reactant in this equation contains magnesium mg and a hydroxide ion oh 1 which is a polyatomic anion

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web nov 9 2023 to balance a chemical equation follow these four simple steps to show the reactants and products write the imbalanced equation calculate the number of atoms of each element on each side of the reaction arrow to make the number of atoms of each element the same on both sides of the equation multiply coefficients the numbers in

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web oct 13 2023 this means that you will need to balance the carbon atoms first 5 use a coefficient to balance the single carbon atom add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation c 3 h 8 o 2 h 2 o 3 co 2