

Name: _____ Date: _____ Class: _____

- 1 What kind of **reaction** has occurred in this picture?

- A synthesis
- B decomposition
- C single replacement
- D double replacement



- 2 In order for a **chemical change** to take place, _____.

- A products must turn into reactants
- B reactants must turn into products
- C heat must be given off
- D heat must be added



- 3 What **type** of **chemical reaction** is shown in the diagram below?



- A double replacement
- B synthesis
- C decomposition
- D single replacement

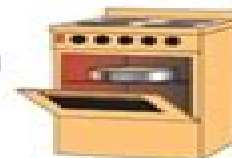
- 4 What is the **evidence** in this picture that a **chemical reaction** has occurred?

- A light and heat
- B sound
- C change of shape
- D change in mass



- 5 Baking a cake is an endothermic reaction that chemically changes the batter into a cake. For an **endothermic reaction** to occur, _____.

- A energy must be released
- B energy must be used
- C shaking is necessary
- D no reactants are used



- 6 Which statement is correct for the **reaction** shown below?

- A it is a decomposition reaction
- B it is a synthesis reaction to make hydrochloric acid
- C mixing certain acids and bases makes salt and water
- D only a physical change has occurred



- 7 What is one way of **speeding up** any chemical reaction?

- A take away the products
- B reduce the amount of reactants
- C increase the amount of reactants
- D decrease the temperature



- 8 What **type** of reaction is the chemical reaction shown below?



- A double replacement
- B single replacement
- C synthesis
- D decomposition

Teachers Manual The Chemical World Activities And Explorations

Michal Rosen-Zvi



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