Dissolving Salt in Water

VALID ARGUMENT FOR BOTH SIDES

Chemical Change

Dissolving salt may be written as a chemical equation. New products are formed, so a chemical reaction occurs.

 $NaCl(s) \rightarrow Na^{+}(aq) + Cl^{-}(aq)$



Physical Change

Salt doesn't react
with the solvent
(water), so no
chemical reaction
occurs. If you remove
the solvent, the
original compound
remains.

Slow Dissolve

Tao Wei



Slow Dissolve Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Slow Dissolve**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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