

5.3.3 Stereo-regular Polymers

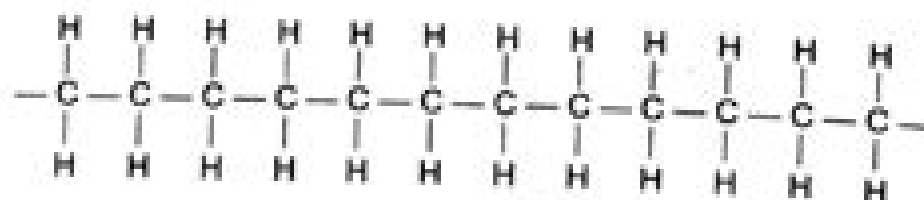
In stereo-regular polymers, each monomer segment is in a regular configuration, giving a definite structural regularity to the polymer molecule as a whole. This structural regularity, in a polymer may arise due to what are termed as "optical" and "geometric" isomerisms of the main chain atoms or substituents in the polymer molecule. The meaning of these two isomerisms is now explained.

Optical Isomerism

Optical isomerism has its origin in the way different substituents sit on an asymmetric carbon atom in a polymer molecule. We will give an example. A polyethylene molecule has fully saturated carbon atoms with the following chemical formula:



We can write this formula in a linear structural way, showing how various atoms are attached (but without bothering about the bond angles and molecular plane) as follows:



Let us now assume that the carbon-carbon bonds are all on the plane of this paper and that the bonds between the carbon atom and its substituent atoms are either below or above the plane of the paper. The structure of polyethylene in a planar zig-zag configuration is then as shown in Fig. 5.6. In doing this, of course, we have not bothered about the free rotation around each carbon-carbon single bond and the various resultant spatial positions of the polymer segment. It is seen from Fig. 5.6 that all carbon atoms are attached to each other through single bonds which are arranged in a zig-zag manner considering their bond angles. All carbon atoms lie in the plane of the paper but hydrogen atoms

Stereoregular Polymers Stereospec Volume

Tobias Bleicker



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