



Smart Start Course **Fact Sheet**

The Office of Adult Education has primary responsibility for implementation of the Mississippi Works Smart Start Career Pathway Model through the Smart Start Pathway Course.

What is the Smart Start Pathway Course?

This course prepares participants for middle-skill level employment. Participants will develop job skills needed for their careers, learn and practice good work habits, and effective communication necessary for successful employment.

- **Why is this course important for participants?** Participants have the opportunity to earn a Smart Start Credential which includes a National Career Readiness Certificate (NCRC). The NCRC demonstrates to employers the participant has the skills needed to be successful in a job. Participants will complete the WorkKeys tests for Applied Math, Graphic Literacy and Workplace Documents. Participants will also learn new skills for future careers and increase their ability for middle-skill level jobs which results in higher pay.
- **How is the course offered?** Smart Start Pathway Courses can be taken face-to-face, online or both face-to-face and online.
- **What are the requirements for participating in a Smart Start Pathway Course?** Participants must be registered in MS WORKS to participate in classes.
- **What is the cost?** This course is free.
- **What are the minimum requirements for achieving a MS Smart Start Credential?**

**Smart
Start** ▶



EDUCATIONAL SKILLS

- **ACTWorkKeys®Curriculum**
 - Applied Math
 - Graphic Literacy
 - Workplace Documents

EMPLOYABILITY SKILLS

- **Career Awareness** (ex: Career Goals, Job Search Strategies, Financial Literacy)
- **Necessary Skills** (ex: Communication & Teamwork, Problem-Solving & Critical Thinking, Employer Expectations)

ACTWorkKeys® RESULTING IN:

- **National Career Readiness Certificate (NCRC)** - Minimum Bronze Level even though Silver is encouraged for increased job opportunities.

Smartstart Your Mississippi Business

Xiang Xie



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they are the same as basic solutions

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NH_2^- and S^{2-} are CH_4 NH_3 and HS^- respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that S^{2-} is the weakest base

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