

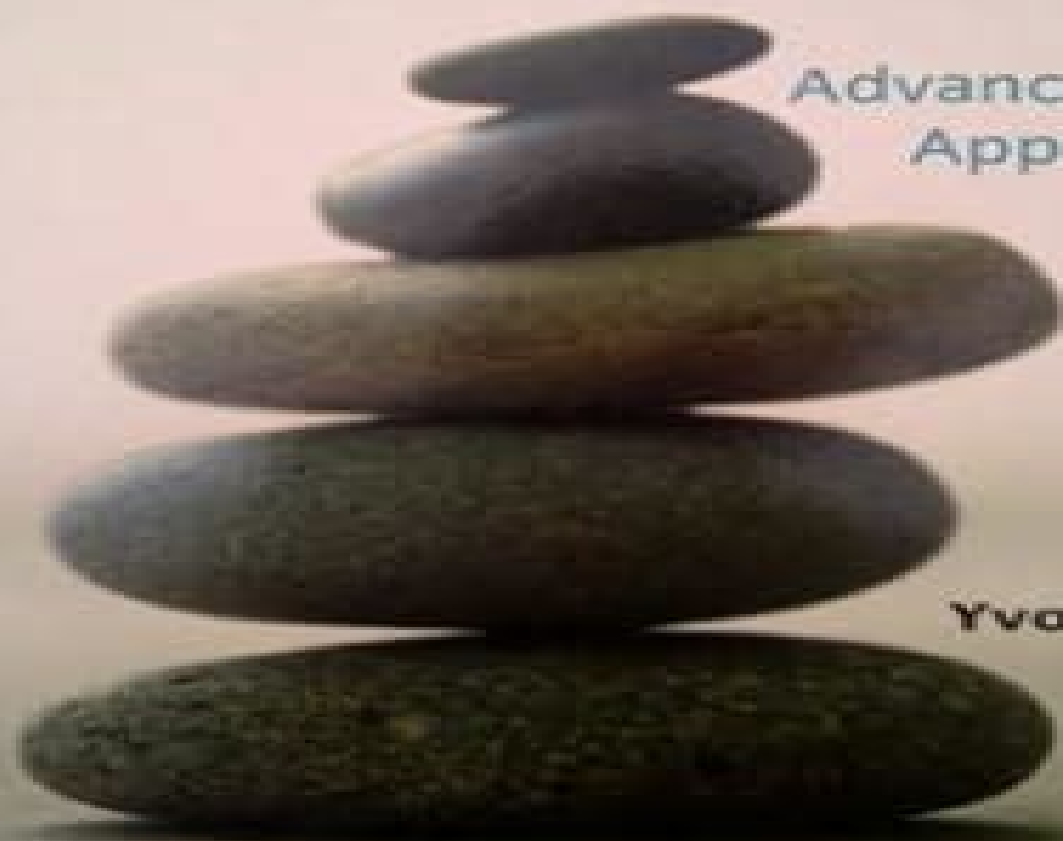
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3

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6 50 g Al $\times \frac{1\text{ mol Al}}{26.98\text{ g Al}} \times \frac{2\text{ mol Al}_2\text{O}_3}{4\text{ mol Al}} = 0.125\text{ mol Al}_2\text{O}_3$

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