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266 section 01 introduction in this lab we observed the relationship between pressure and temperature

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of

declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95

82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates

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