Activity No.2

SPECIFIC HEAT OF SOLIDS

INTRODUCTION

Calorimetry means "measuring heat." When heat flow occurs between two bodies that are isolated from their surroundings, the amount of heat lost by one body must be equal to the amount of heat gained by the other. Heat is a form of energy in transit, so this principle is just a conservation of energy. We take each quantity added to a body as positive and each quantity leaving a body as negative. When several bodies interact, the algebraic sum of the quantities of heat transferred to all the bodies must be zero.

The mathematical representation of the amount of heat Q needed to achieve a temperature change by a material with mass m and specific heat capacity c is

where T = final temperature, and initial temperature.

In an isolated system, there is no transfer of either mass or energy to its environment. For such case the total heat in the system is conserved, that is,

Specific Heat Capacity for Some Materials/Substance

Material/Substance	Specific Heat c	
	(cal/g C°)	(J/Kg C°)
Water	1.000	4190
Copper	0.093	390
Lead	0.031	130
Steel	0.118	470
Aluminum	0.217	910

OBJECTIVE

To measure the specific heat capacity of two kinds of solid specimens by calorimetry.

APPARATUS

Steam generator with cup, calorimeter, 2 thermometers, 2 kinds of metal shots, platform balance, slotted masses, 600 ml. beaker, electric stove, extension cord, rug

PROCEDURE

https://www.youtube.com/watch?v=d4iCJakz2i8

 Fill the steam generator 1/2 full of water and start heating it over the stove. Set the stove range dial to medium.

Record the two kinds of metal shots in Table 2.1.

3. Weigh about 100 grams of the first metal shots. Pour these in the steam can (inner cup of the steam generator) and place the can in the steam generator. Cover the steam can with a cork with two holes, one for the thermometer and the other for the stirrer. Heat the shots over the steam generator and stir occasionally. Do not use the thermometer to stir the shots but keep the thermometer embedded in the shots. Continue heating and observe the thermometer until all the

Specific Heat Of Solids

Rachel S Tattersall

Specific Heat Of Solids:

Whispering the Techniques of Language: An Mental Journey through Specific Heat Of Solids

In a digitally-driven world where displays reign great and instant conversation drowns out the subtleties of language, the profound techniques and psychological nuances concealed within words often go unheard. However, situated within the pages of **Specific Heat Of Solids** a charming literary value sporting with raw feelings, lies an exceptional journey waiting to be undertaken. Published by a skilled wordsmith, this marvelous opus encourages readers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the very material of every word. Within the emotional depths of this touching review, we can embark upon a heartfelt exploration of the book is primary styles, dissect its interesting writing type, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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Table of Contents Specific Heat Of Solids

- 1. Understanding the eBook Specific Heat Of Solids
 - The Rise of Digital Reading Specific Heat Of Solids
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Specific Heat Of Solids
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Specific Heat Of Solids
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Specific Heat Of Solids
 - Personalized Recommendations
 - Specific Heat Of Solids User Reviews and Ratings

- Specific Heat Of Solids and Bestseller Lists
- 5. Accessing Specific Heat Of Solids Free and Paid eBooks
 - Specific Heat Of Solids Public Domain eBooks
 - Specific Heat Of Solids eBook Subscription Services
 - o Specific Heat Of Solids Budget-Friendly Options
- 6. Navigating Specific Heat Of Solids eBook Formats
 - o ePub, PDF, MOBI, and More
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 - Specific Heat Of Solids Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Specific Heat Of Solids
 - Highlighting and Note-Taking Specific Heat Of Solids
 - Interactive Elements Specific Heat Of Solids
- 8. Staying Engaged with Specific Heat Of Solids
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Specific Heat Of Solids
- 9. Balancing eBooks and Physical Books Specific Heat Of Solids
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Specific Heat Of Solids
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Specific Heat Of Solids
 - Setting Reading Goals Specific Heat Of Solids
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Specific Heat Of Solids
 - Fact-Checking eBook Content of Specific Heat Of Solids
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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