

Physical Chemistry Formula Revision

Basic Concepts

$$\text{Average atomic mass} = \frac{a_1x_1 + a_2x_2 + a_3x_3 + \dots}{100}$$

 a_1, a_2, a_3, \dots = atomic mass of isotopes. x_1, x_2, x_3, \dots = percentage abundance

Model concept

$$\text{na. of males} = n = \frac{\text{given mass}}{\text{avg. mass}} = \frac{M}{m}$$

$$n = \frac{\text{no. of particles}}{\text{Avogadro's no.}} = \frac{N}{6.022 \times 10^{23}}$$

$$n = \frac{\text{vol. of gas}}{\text{vol. of gas at STP}} = \frac{V}{22.4 \text{ l}}$$

Concentration Times

$$\text{Percentage, \%} = \frac{W_b \text{ or } V_b}{W \text{ or } V} \times 100$$

$$\text{Molarity, } M = \frac{n_b}{V(l)} = \frac{N_b \times 1000}{M_b \times V(ml)}$$

$$\text{Normality, } N = \frac{m_{\text{eq}}}{V(l)} = \frac{M_b \times 1000}{E_b \times V(\text{ml})}$$

$$\hookrightarrow \frac{M_b}{n_f}$$

$$\text{parts per million, ppm} = \frac{W_b}{W} \times 10^6$$

$$\text{Molality, } m = \frac{n_b}{M_b (\text{kg})} = \frac{M_b \times 1000}{M_b \times M_b (\text{g})}$$

Mole fraction, $x_b = \frac{n_b}{n_a + n_b + n_c} \dots \dots$

Stoichiometry



$$\frac{\text{moles of A}}{a} = \frac{\text{moles of B}}{b} = \frac{\text{moles of C}}{c}$$

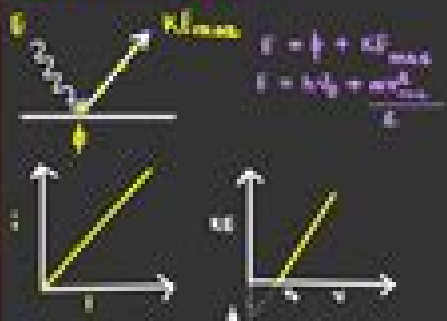
Limiting R_p : $\frac{\text{given amt.}}{\text{stoichiometric coeff.}} \rightarrow \text{least}$

1. **Introduction**

Plankton Greenhouse Warming

$$E = \frac{hc}{\lambda} \quad c(\text{Å}) = \frac{12400}{\lambda(\text{Å})}$$

Positive External Effects



The University of Maryland

$$m_e \lambda = \frac{h}{\lambda} \rightarrow \lambda = \frac{h}{m_e v}$$

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$$\sigma_{\text{max}} = \frac{m \cdot \omega^2}{1 - \omega^2} \quad T = 0.15 \times \frac{m}{L} \text{ s}$$

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$$E_{\text{cell}} = E_{\text{cell}}^{\circ} - \frac{RT}{nF} \ln Q \quad \text{where } Q = \frac{[\text{H}_2\text{O}][\text{H}_2]}{[\text{H}_2\text{O}_2]}$$

$$\frac{1}{\lambda} = \frac{1}{\lambda_0} \left[\frac{1}{\sqrt{1 - \frac{v^2}{c^2}}} - 1 \right] \quad \text{for } v \ll c$$

more on of spectral lines where λ varies from λ_1 to λ_2 :
$$\frac{1}{\lambda_2 - \lambda_1} \int_{(\lambda_1 - \lambda_2)}^{\lambda_2 - \lambda_1} \frac{1}{\lambda} d\lambda$$

Heisenberg's Uncertainty Principle: $\Delta x \Delta p \geq \frac{h}{4\pi}$

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total nodes = 100

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$$\text{modified_nodes} = m - 1 - l$$

Keywords: *work engagement, organizational commitment, turnover intentions, organizational citizenship behaviors, organizational identification*

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$\frac{1}{2} \times \frac{1}{2} = \frac{1}{4}$

Structure Of Physical Chemistry

C. N. (Cyril Norman Hinshelwood (Sir))



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the presentation assists students in developing an intuitive understanding of the subjects as well as skill in quantitative manipulations The unifying nature of physical chemistry is emphasized in the book by its organization beginning with atoms and molecules and proceeding to molecular assemblies of increasing complexity ending with the emergence of matter that carries information i e the origin of life a physicochemical process of unique importance The aim is to show the broad scope and coherence of physical chemistry **General Catalogue** Massachusetts Institute of Technology,1927 The Annual Catalogue of Purdue University, Lafayette, Indiana ... with Announcements for ... Purdue University,1916 Academic Press Dictionary of Science and Technology Christopher G. Morris,Academic Press,1992-08-27 A Dictionary of Science and Technology Color Illustration Section Symbols and Units Fundamental Physical Constants Measurement Conversion Periodic Table of the Elements Atomic Weights Particles The Solar System Geological Timetable Five Kingdom Classification of Organisms Chronology of Modern Science Photo Credits **The Ceramist** ,1926 Annual Report National Research Council of Canada,1927 *Western Reserve University. School of Education. Afternoon and Evening Classes* Western Reserve University. School of Education,1929

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