

Period 2

P-block

CAS Number 7440-42-8

Allotropes α -, β -rhombohedral, γ -rhombohedral (and more)

Oxidation states -5, -1, 0, +1, +2, +3 (mainly with acids)

Boiling point 4000 K (3727 °C, 6741 °F)

Natural occurrence primordial

Group 13 (boron group)

Molar heat capacity 10.87 J/(mol·K)

Thermal expansion β form: 5-7 $\mu\text{m}/(\text{m}\cdot\text{K})$ at 25 °C

Melting point 2349 K (2076 °C, 3767 °F)

Ionization energies

1st 800.6 kJ/mol

2nd 2427.1 kJ/mol

3rd 3658.7 kJ/mol

(more)

Density when liquid (at m.p.) 2.08 g/cm³

Standard atomic weight A_r , std(B) [10.806, 10.83] conventional 10.8

Speed of sound (at m.p.) 6100 m/s at 25 °C

Electrons per shell 2, 3

BIG

Atomic radius empirical 90 pm

Element category Metalloid

Phase at STP solid

Van der Waals radius

BIG

Electronegativity Pauling scale: 2.04

Appearance black-brown

Crystal structure rhombohedral

A	III	B
2	pin B Boron $2s^2 2p^1$ 10.806	5 3 2

The Analysis Of Elemental Boron

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W. Hovey, Nick Derick, P. T. Goin, Sabri Ergun, 1961

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Table of Contents The Analysis Of Elemental Boron

1. Understanding the eBook The Analysis Of Elemental Boron
 - The Rise of Digital Reading The Analysis Of Elemental Boron
 - Advantages of eBooks Over Traditional Books
2. Identifying The Analysis Of Elemental Boron
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an The Analysis Of Elemental Boron
 - User-Friendly Interface
4. Exploring eBook Recommendations from The Analysis Of Elemental Boron
 - Personalized Recommendations
 - The Analysis Of Elemental Boron User Reviews and Ratings
 - The Analysis Of Elemental Boron and Bestseller Lists
5. Accessing The Analysis Of Elemental Boron Free and Paid eBooks

- The Analysis Of Elemental Boron Public Domain eBooks
- The Analysis Of Elemental Boron eBook Subscription Services
- The Analysis Of Elemental Boron Budget-Friendly Options
- 6. Navigating The Analysis Of Elemental Boron eBook Formats
 - ePub, PDF, MOBI, and More
 - The Analysis Of Elemental Boron Compatibility with Devices
 - The Analysis Of Elemental Boron Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of The Analysis Of Elemental Boron
 - Highlighting and Note-Taking The Analysis Of Elemental Boron
 - Interactive Elements The Analysis Of Elemental Boron
- 8. Staying Engaged with The Analysis Of Elemental Boron
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers The Analysis Of Elemental Boron
- 9. Balancing eBooks and Physical Books The Analysis Of Elemental Boron
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection The Analysis Of Elemental Boron
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine The Analysis Of Elemental Boron
 - Setting Reading Goals The Analysis Of Elemental Boron
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of The Analysis Of Elemental Boron
 - Fact-Checking eBook Content of The Analysis Of Elemental Boron
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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