

The term stripping stands for the fact that, during the determination, the accumulated analyte is removed from the working electrode. This process can be followed voltammetrically, potentiometrically, or chronopotentiometrically, this is expressed by the terms stripping voltammetry, stripping potentiometry, and stripping chronopotentiometry.

Accumulation always takes place at constant potential (E_{acc} , accumulation potential) at a stationary mercury drop, mercury film, graphite or noble metal electrode for a controlled period (t_{acc} , accumulation time). The analyte is deposited electrolytically as a metal, as a sparingly soluble mercury compound or adsorptively as a complex compound.

Stripping Potentiometry Novel Methodology And Electrode Design

Yudong Wang



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