

# Solid Liquid Equilibrium



The *melting point* of a solid or the *freezing point* of a liquid is the temperature at which the solid and liquid phases coexist in equilibrium.

**Molar heat of fusion** ( $\Delta H_{\text{fus}}$ ) is the energy required to melt 1 mole of a solid substance at its freezing point.

$$\Delta H_{\text{fus}} = -\Delta H_{\text{solid}}$$

# Solid Liquid Equilibrium

**Clemens Wendtner**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

## **Solid Liquid Equilibrium:**

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