Solid Liquid Equilibrium

$$H_2O(s) \longrightarrow H_2O(l)$$

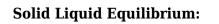
The melting point of a solid or the freezing point of a liquid is the temperature at which the solid and liquid phases coexist in equilibrium.

Molar heat of fusion (ΔH_{fus}) is the energy required to melt 1 mole of a solid substance at its freezing point.

$$\Delta H_{\text{fus}} = -\Delta H_{\text{solid}}$$

Solid Liquid Equilibrium

Clemens Wendtner



Uncover the mysteries within Crafted by is enigmatic creation, Embark on a Mystery with **Solid Liquid Equilibrium**. This downloadable ebook, shrouded in suspense, is available in a PDF format (Download in PDF: *). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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