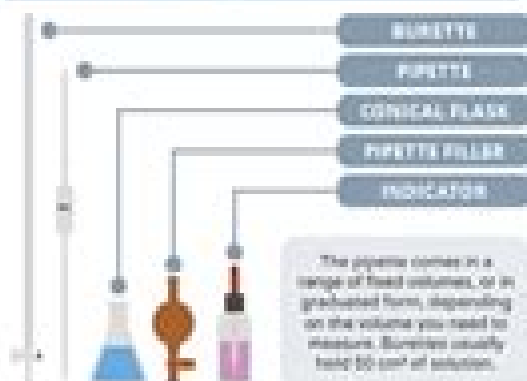


# CHEMISTRY TECHNIQUES: TITRATION

Used to determine the concentration of a particular solution, by measuring how much of a solution of known concentration reacts with a known volume of it.

## EQUIPMENT



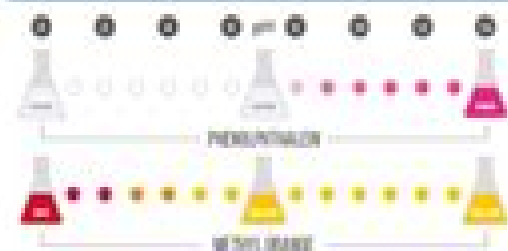
## USING A PIPETTE



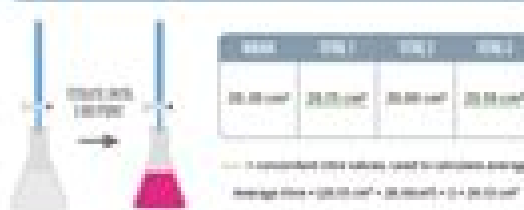
## FILLING THE BURETTE



## USING INDICATORS

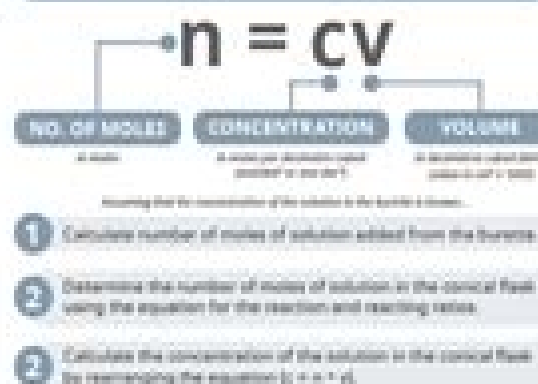


## CARRYING OUT THE TITRATION



To carry out the titration, the tip of the burette is opened to allow the solution inside to flow into a known volume of the solution in the conical flask. The amount of solution from the burette required to reach the end point is recorded. A rough titration is usually followed by more accurate runs. Multiple titrations are carried out until concordant titres are obtained (within 0.10 cm<sup>3</sup> of each other).

## CARRYING OUT CALCULATIONS



# Techniques Of Chemistry

**Rachel Sandford**



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