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# The Harvard College Guide To Grants

**Peter Vogt**



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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below normal boiling point of  $\text{NH}_3$  33.3 °C  $\Delta H_{\text{vap}}$  23.4 kJ/mol specific heat  $\text{NH}_3$  1.474 J/g·K specific heat  $\text{NH}_3$  2.06 J/g·K

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