

# SMALL COMFORTS



More Comments and Comic Pieces by

## TOM BODETT

# Small Comforts More Comments And Comic Pieces

**Tom Bodett**



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this video accompanies the chm 252 lab nitration of bromobenzene in this video students will get an introduction to the lab as well as the set up for conducting the reaction students should

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction e.g. particularly nitration of methyl benzoate to give methyl m-nitrobenzoate thus determine the weight, melting point and percentage yield of the pure methyl m-nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene. Bromine substituted on an aromatic ring is an ortho/para directing group. It will direct an incoming electrophile to the ortho and para positions. It is also deactivating, meaning that its reactions will be slower than the identical reactions of benzene.

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1. Go to the restroom and wash the affected area with lots of soap and water. The soap in the lab is not suitable for this purpose. 2. Return to the lab and apply mineral oil to the affected area. 3. The summary to this warning is not to touch any part of your body while performing this experiment. If you must leave the lab for any reason, 1.

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In the nitration of bromobenzene, a mixture of sulfuric acid and nitric acid reacts with one another. Nitric acid, the weaker of the two acids, will serve as a base to form the nitronium ion. Bromobenzene will attack the nitronium ion by nucleophilic attack.

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Then actual is your actual findings yield actual theoretical  $\times 100$ . What are the possible side reactions of nitration of bromobenzene? Dinitration can occur at ortho/para or at both ortho positions. Overheating meta product happens very rarely because bromine is an ortho/para director.

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explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture  $\text{HNO}_3$  3 M  $\text{H}_2\text{SO}_4$  in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion  $\text{NO}_2^+$  which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o-bromonitrobenzene

and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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web ray made his directorial debut in 1955 with pather panchali and directed 36 films comprising 29 feature films five documentaries and two short films although ray s work generally received critical acclaim 5 6 his film pather panchali and ashani sanket 1973 were criticised for exporting poverty and distorting india s image abroad

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