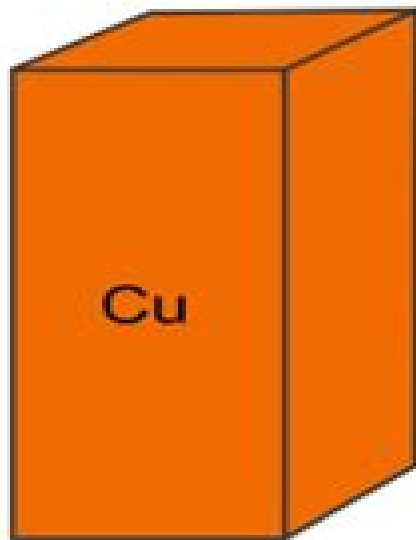
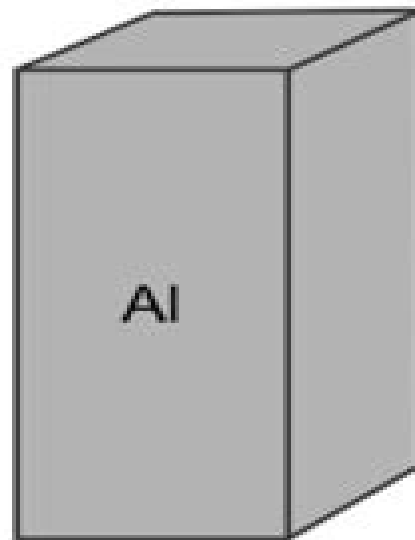




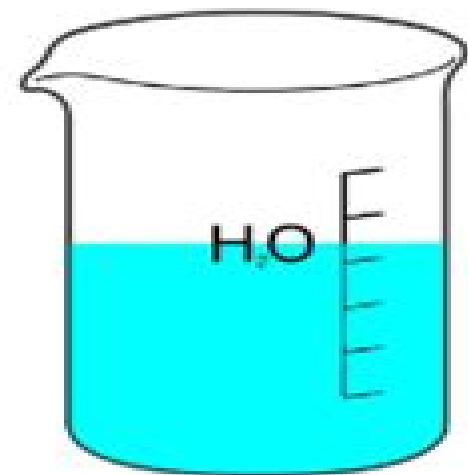
copper block

specific heat  
of copper  
 $= 390 \text{ J/kg}^\circ\text{C}$



aluminum block

specific heat  
of aluminum  
 $= 910 \text{ J/kg}^\circ\text{C}$



water

specific heat  
of water  
 $= 4200 \text{ J/kg}^\circ\text{C}$



Lower Specific Heat Capacity -

warms up and cools down quickly  
as it takes much less energy to  
change its temperature

Higher Specific Heat Capacity -

warms up and cools down slowly  
as it takes up much more energy  
to change its temperature

# Specific Heat Of Solids

**Zhe Xu**



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