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**Bruce Levine**



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web jun 2 2020 the reactant that is consumed first and limits the amount of products that can be obtained is the limiting reactant to identify the limiting reactant calculate the number of moles of each reactant present and compare this ratio to the mole ratio of the reactants in the balanced chemical equation

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web reactant present in an amount lower than required by the reaction stoichiometry thus limiting the amount of product generated molecular equation chemical equation in which all reactants and products are represented as neutral substances

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web the key to recognizing which reactant is the limiting reagent is based on a mole mass or mass mass calculation

whichever reactant gives the lesser amount of product is the limiting reagent what we need to do is determine an amount of one product either moles or mass assuming all of each reactant reacts

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web they have ribosomes but these are not enclosed in a membrane eukaryotic cells contain various membrane bound organelles including mitochondria endoplasmic reticulum golgi apparatus lysosomes and more size prokaryotic cells generally smaller with typical sizes ranging from 0.1 to 5 micrometers

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