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Tether Poems

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although the c-cl bonds are rather polar the individual bond dipoles cancel one another in this symmetrical structure and does not have a net dipole moment

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 determine based on electronegativity and symmetry if the following molecules are polar or nonpolar
 hcn h₂c=cf₃ c₂f₄ c₂h₂cl₂ c₂o c₂l₂ nf₃ n₂f₄ q3

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 a. clf₅ b. clo₂ c. tecl₂ d. pcl₃ e. sef₄ f. ph₂ g. xef₂ answer problem 6.2.3.6.2.3

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web the relative electronegativities of the bonded atoms is h < c < cl and so the bond moments all point toward the cl end of the

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