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Solutions, Minerals, and Equilibria Robert Minard Garrels, Charles Louis Christ, 1990-01-01 **Solutions, Minerals, and Equilibria** Robert Minard Garrels, 1967 SOLUTIONS, MINERALS & EQUILIBRIA. GARRELS., 1954 *Uranium Solution-mineral Equilibria at Low Temperatures with Application to Sedimentary Ore Deposits* Donald Langmuir, 1977

Phosphate Minerals J.O. Nriagu, P.H. Moore, 2012-12-06 The literature on the geology chemistry and biochemistry of phosphorus generally takes its mineralogy for granted The incidental information on phosphate minerals given in these texts is often obsolescent and inaccurate The few mineralogical texts that have dealt comprehensively with the phosphate minerals have now become outdated and typically present the essential information in a manner unsuitable for nongeological readers This volume is intended as a ready reference for workers who require good basic information on phosphate minerals or their synthetic equivalents The topics covered should appeal to geologists and geochemists lithologists environmental scientists and engineers chemists and biochemists who have any interest in the intricate world of phosphorus The hard tissues of many vertebrates and the many pathological calcifications consist mostly of phosphate minerals The precipitation of these compounds also plays a major role in the ecological cycling of phosphorus and occasionally even dominates the behavior of many trace metals in many geochemical and biological systems Indeed many pegmatitic phosphate minerals have acquired some notoriety because of the rarer trace metals which they tend to accumulate With the commercialization of phosphate fertilizers since the early part of the 19th century phosphate minerals have assumed an important role in industrial chemistry and agriculture Clearly the study of phosphate minerals is important from the economic agricultural environmental and human and animal health viewpoint Thermodynamic Properties of Minerals and Related Substances at 298.150 K (25.00 C) and One Atmosphere (1.013 Bars) Pressure and at Higher Temperatures Richard A. Robie, David R. Waldbaum, 1968

Thermodynamic Properties of Minerals and Related Substances at 298.15° K (25.0° C) and One Atmosphere (1.013 Bars) Pressure and at Higher Temperatures Richard A. Robie, David R. Waldbaum, 1968 A summary of the thermodynamic data for minerals at 298.15 K together with calculated values of the functions H_0^f , $T G_0^f$, $T S_0^f$ and G_0^f at temperatures up to 2000 K **Strategic and Critical Minerals in the Midcontinent Region, United States** Warren C. Day, Diane E. Lane, 1992 **Geochemistry and Mineral Formation in the Earth Surface** Rafael Rodríguez-Clemente, Yves Tardy, 1987 Water-resources Investigations, 1980

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