Due Friday, January 30

Problem Set #1

Report each answer with proper significant figures in mind.

1. Succinic acid dissociates as follows:

$$H_2C_4H_4O_4 + H_2O \Rightarrow HC_4H_4O_4^- + H_3O^+$$
 $K_{31} = 6.2 \times 10^{-5}$

$$HC_4H_4O_4^- + H_2O \Rightarrow C_4H_4O_4^{2-} + H_3O^+$$
 $K_{12} = 2.3 \times 10^{-6}$

Calculate K61 and K62 for the following reactions (3 points):

$$C_4H_4O_4^{2-} + H_2O \Rightarrow HC_4H_4O_4^{-} + OH^{-}$$
 $K_{01} = ?$

$$HC_4H_4O_4^- + H_2O \Rightarrow H_2C_4H_4O_4 + OH^ K_{02} = ?$$

(hint: you need another reaction other than those listed to do this)

Solving Problems In Analytical Chemistry

G Thomas

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