

# Alkali Metals



**Sodium**



**Lithium**



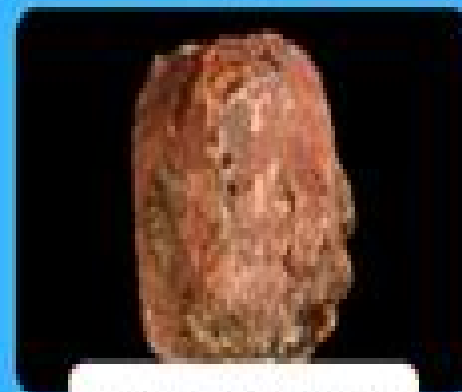
**Potassium**



**Rubidium**



**Cesium**



**Francium**

# Sodium And The Alkali Metals

**Hans H. Ussing, Poul Kruhoffer, Hess J.  
Thaysen, N.H. Thorn**

## Sodium And The Alkali Metals:

**Sodium and the Alkali Metals** Nigel Saunders, 2004 Discusses the origin discovery special characteristics and uses of the sixth most abundant element in the Earth's crust sodium and the other group one metals **Sodium and the Alkali Metals** Nigel Saunders, 2004-08-01 This series uses a common or well known element to look at the groups of the Periodic Table Each book will be introduced by the full periodic table and then broken down into the group being covered

Characteristics uses and histories of individual elements are explored **The Alkali Metals** Kristi Lew, 2009-08-15 The alkali metals are a fascinating group of six elements that are necessary to life on Earth The Alkali Metals explores the original discovery of these six elements how they are classified on the periodic table compounds uses and how these elements impact the reader individually *Periodic Table: Sodium and the Alkali Metals / Aluminium and the Group 3 Elements / Nitrogen and the Group 5 Elements / Fluorine and the Halogens / Hydrogen / Uranium* Nigel Saunders, 2004-02-05

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web the atomic mass of a single atom is simply its total mass and is typically expressed in atomic mass units or amu by definition an atom of carbon with six neutrons carbon 12 has an atomic mass of 12 amu other atoms don t generally have round number atomic masses for reasons that are a little beyond the scope of this article

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web jul 29 2021 as you learned the mass number is the sum of the numbers of protons and neutrons present in the nucleus of an atom the mass number is an integer that is approximately equal to the numerical value of the atomic mass although the mass number is unitless it is assigned units called atomic mass units amu because a molecule or a

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web atomic structure google classroom what three particles make up an atom choose 1 answer protons neutrons and photons a protons neutrons and photons positrons neutrons and electrons b positrons neutrons and electrons protons electrons and positrons c protons electrons and positrons protons neutrons and electrons d