Solidification Structures of Aluminum Alloys

Doru M. Stefanescu and Roxana Ruxanda. The University of Alabama

ALUMINUM-BASE ALLOYS constitute a group of cast materials second only to ferrous castings in tonnage. Aluminum alloys have gained increasing markets in the automotive and airceaft industry. New alloys are still under development, since there are still wide possibilities. of improving their properties by using simple, affordable techniques, such as alloying and heat treating. A wide range of materials can be added to aluminum (Ref 1): zinc, magnesium, copper, silicon, iron, lithium, manganese, nickel, silver, tin, and titanium. The solid solubility of these elements in aluminum varies considerably. Some elements are used as solid-solution strengtheners, others form desirable intermetallic compounds. The most common alloy systems are aluminum-silicon, aluminum-copper, and aluminum-magnesium. Some typical solidification microstructures for these alloying systems are given in the following sections.

Basic Microstructures of Aluminum-Base Alloys

Aluminum-Silicon Alloys. The equilibrium phase diagram of the binary aluminum-silicon system is similar to that in Fig. 16(b) of the article. "Fundamentals of Solidification" in this Volume, representing partial solid solubility with a cutectic reaction. The cutectic invariant is at 11.7% Si and 577 °C (1070 °F). Typical examples of hypocutectic, cutectic, and hypercutectic aluminum-silicon commercial alloys are given in Fig. 1 (Ref 2).

Binary eutectic or hypocutectic aluminum-silicon alloys are characterized by good castability and corrosion resistance. However, the aluminum-silicon alloys are seldom binary. Strengthening of aluminum-silicon alloys is achieved by adding small amounts of other elements, such as copper, magnesium, or iron. Higher iron contents can promote the formation of britle plates of AlFeSi or other complex intermetallies in the presence of manganese. This may have a negative influence on the mechanical properties.

Aluminum-Copper Alloys. The equilibrium diagram of the binary aluminum-copper system is given in Fig. 2. These alloys are typical solid-solution alloys generally containing 4 to 6% Cu and some magnesium. After a solution treatment, CuAl₂ particles precipitate from the quenched alloy, as seen in Fig. 2. The level of impurities can affect both the macrostructure and

microstructure. This is demonstrated in Fig. 3 and 4 for a nominal Al-4.5Cu-0.25Mg alloy with a small amount of silver (0.7%) and titanium, respectively. Titanium addition considerably decreases the grain size.

Aluminum-Magnesium Alloys. The corner of the phase diagram of practical interest is similar to that of aluminum-copper alloys, with the maximum solubility of magnesium in aluminum at 17.4% Al. Some typical examples of aluminum-magnesium commercial alloys are given in Fig. 5.

Other Multicomponent Aluminum Alloys, Investigations of higher-order multicomponent alloys solidification are rather limited. Sophisticated methods are used to identify phases and their orientation. An example of the microstructure of the Al 7050 alloy containing 11 elements is given in Fig. 6 (Ref 3). The dark areas are qualuminum phase and the bright areas are qualernary sigma (of) phase (Al,Cu,Zn),Mg, tensary S phase (Al,MgCu), or/and binary 0 phase (Al,MgCu). These phases were identified using optical inetallography, scanning electron microscopy (SEM), x-ray map, and electron probe microsandysis (EPMA) techniques. The dendrite arms are separated by sigma, S, B, and/or eutre-





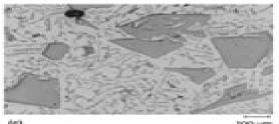


Fig. 1 Typical micrositructures of Improveducitic, entercitic, and hypometeritic aluminum-solicon commercial alloys, tall Hypometeritic aluminum-validom alloy 6A-5.7%, alloying 6A-11%, alloying a respective for the complex extension with the surface of the complex extension. Behavior of the Conginual magnification 110 X, all Datestic aluminum-validom alloy 6A-11.8%, alloy type A3-9%, a subminum dendrition, primary allocate particles, and areas with complex extension. Behavior 0.5%, 4B. Original magnification 1.80 X, is 1.85 personal extension alloy 6A-15%, alloy type A3-9%. Large primary efficient particles, extension crystals, and A1-Cu ghave in a matrix of a subminum phase. Behavior, 0.5%, 4B. Original magnification 1.10 X.

Solidification Of Aluminum Alloys

Alexander Vorozhtsov

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rodnih i društvenih pojava prirodnih nauka i društvenih disciplina 1 $3\,$