



Small Game

Mary C. Stiner, Ofer Bar-Yosef



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 $\text{Pb} + \text{OH}^- + 2\text{HCl} \rightarrow \text{H}_2\text{O} + \text{PbCl}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 + 6\text{KBr} \rightarrow 2\text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2$
 $\text{CO}_2 + 2\text{H}_2\text{O} \rightarrow \text{C}_3\text{H}_8 + 5\text{O}_2$
 $3\text{CO}_2 + 4\text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 25\text{O}_2$
 $16\text{CO}_2 + 18\text{H}_2\text{O}$

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 gaseous butane C_4H_{10} reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$
 $\text{Mg} + \text{F}_2 \rightarrow 2\text{MgF}_2$
 $\text{Li}_2\text{CO}_3 + \text{MgCO}_3 \rightarrow \text{LiF} + \text{P}_4\text{O}_{10}$
 $2\text{P}_2\text{O}_5 + 3\text{RbNO}_3 \rightarrow \text{BeF}_2 + \text{Be}_2\text{O}_3 + 2\text{RbF}$

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 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 + 6\text{KBr} \rightarrow 2\text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2$
 $\text{CO}_2 + 2\text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 5\text{O}_2$
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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $\text{HNO}_3 + \text{H}_2\text{O} \rightarrow 2\text{O}_2 + 3\text{H}_2\text{O}$
 $2\text{H}_2\text{O} + 2\text{O}_2 + 4\text{RbH} + 2\text{O} + \text{RbOH} + 2\text{H}_2 + 5\text{Cl}_2 + 2\text{KBr} + \text{KCl} + \text{Br}_2 + 6\text{N}_2 + 2\text{H}_2 + 3\text{NH}_3 + 7\text{N}_2\text{F}_6 + 3\text{Ca} + \text{N}_2\text{O} + \text{C}_2\text{H}_5\text{OH}$

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