

Gas	Chemical formula	Molecular mass	R (kJ/kg-K)	ρ (kg/m ³)	C_{p0} (kJ/kg-K)	C_{v0} (kJ/kg-K)	$\gamma = \frac{C_p}{C_v}$
Steam	H ₂ O	18.015	0.4615	0.0231	1.872	1.410	1.327
Acetylene	C ₂ H ₂	26.038	0.3193	1.05	1.699	1.380	1.231
Air	—	28.97	0.287	1.169	1.004	0.717	1.400
Ammonia	NH ₃	17.031	0.4882	0.694	2.130	1.642	1.297
Argon	Ar	39.948	0.2081	1.613	0.520	0.312	1.667
Butane	C ₄ H ₁₀	58.124	0.1430	2.407	1.716	1.573	1.091
Carbon dioxide	CO ₂	44.01	0.1889	1.775	0.842	0.653	1.289
Carbon monoxide	CO	28.01	0.2968	1.13	1.041	0.744	1.399
Ethane	C ₂ H ₆	30.07	0.2765	1.222	1.766	1.490	1.186
Ethanol	C ₂ H ₅ OH	46.069	0.1805	1.883	1.427	1.246	1.145
Ethytlene	C ₂ H ₄	29.054	0.2964	1.138	1.548	1.252	1.237
Helium	He	4.003	2.0771	0.1615	5.193	3.116	1.667
Hydrogen	H ₂	2.016	4.1243	0.0813	14.209	10.085	1.409
Methane	CH ₄	16.043	0.5183	0.648	2.254	1.736	1.299
Methanol	CH ₃ OH	32.042	0.2595	1.31	1.405	1.146	1.227
Neon	Ne	20.183	0.4120	0.814	1.03	0.618	1.667
Nitric oxide	NO	30.006	0.2771	1.21	0.993	0.716	1.387
Nitrogen	N ₂	28.013	0.2968	1.13	1.042	0.745	1.400
Nitrous oxide	N ₂ O	44.013	0.1889	1.775	0.879	0.690	1.274
Oxygen	O ₂	31.999	0.2598	1.292	0.922	0.662	1.393
Propane	C ₃ H ₈	44.094	0.1886	1.808	1.679	1.490	1.126
R-12	CCl ₂ F ₂	120.914	0.06876	4.98	0.616	0.547	1.126
R-22	CHClF ₂	86.469	0.09616	3.54	0.658	0.562	1.171
R-32	CF ₂ H ₂	52.024	0.1598	2.128	0.822	0.662	1.242
R-125	CHF ₂ CF ₃	120.022	0.06927	4.918	0.791	0.721	1.097
R-134a	CF ₃ CH ₂ F	102.03	0.08149	4.20	0.852	0.771	1.106
Sulphur dioxide	SO ₂	64.059	0.1298	2.618	0.624	0.494	1.263
Sulphur trioxide	SO ₃	80.053	0.10386	3.272	0.635	0.531	1.196

^aOr saturation pressure if it is less than 100 kPa.

Tables Of Thermodynamic Properties Of Am

William F. Hanna



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The NBS Tables of Chemical Thermodynamic Properties Donald D. Wagman, 1982 Properties Tables Booklet for Thermal Fluids Engineering Stephen Turns, David Kraige, 2007-08-13 This booklet is an ideal supplement for any course in thermodynamics or the thermal fluid sciences and a handy reference for the practising engineer The tables in the booklet complement and extend the property tables in the appendices to Stephen Turns's Thermodynamics Concepts and Applications and Thermal Fluid Sciences An Integrated Approach In addition to duplicating the SI tables in these books it extends the tables to cover US customary units as well The booklet also contains property data for the refrigerant R 134a and properties of the atmosphere at high altitudes **Thermodynamics For Dummies** Michael Pauken, 2025-03-17 The thermodynamics knowledge you need to succeed in class and in your career Thermodynamics For Dummies 2nd Edition covers the topics found in a typical undergraduate introductory thermodynamic course which is an essential course to nearly all engineering degree programs It also brings the subject to life with exciting content on where and how thermodynamics is being used today spoiler alert everywhere You'll grasp the basics of how heat and energy interact thermodynamic properties of reactions and mixtures and how thermodynamic cycles are used to make things go This useful guide also covers renewable energy systems new refrigerant technology and a more diverse perspective on the history of the field Within you'll Get clear explanations of the laws of thermodynamics thermodynamic cycles and beyond Read about real world examples to help you connect with the content Practice solving thermodynamic problems to internalize what you've learned For students looking for resources to demystify thermodynamics Thermodynamics For Dummies 2nd Edition is the perfect choice Become thermodynamically savvy with this accessible guide *Geological Survey Bulletin* Geological Survey (U.S.), 1978

Weak-field Magnetic Susceptibility Anisotropy and Its Dynamic Measurement William F. Hanna, 1977

Thermodynamics N.A. Gokcen, R.G. Reddy, 2013-11-11 This edition of Thermodynamics is a thoroughly revised streamlined and corrected version of the book of the same title first published in 1975 It is intended for students practicing engineers and specialists in materials sciences metallurgical engineering chemical engineering chemistry electrochemistry and related fields The present edition contains many additional numerical examples and problems Greater emphasis is put on the application of thermodynamics to chemical materials and metallurgical problems The SI system has been used throughout the textbook In addition a floppy disk for chemical equilibrium calculations is enclosed inside the back cover It contains the data for the elements oxides halides sulfides and other inorganic compounds The subject material presented in chapters III to XIV formed the basis of a thermodynamics course offered by one of the authors R G Reddy for the last 14 years at the University of Nevada Reno The subject matter in this book is based on a minimum number of laws axioms and postulates This procedure avoids unnecessary repetitions often encountered in books based on historical sequence of development in thermodynamics For example the Clapeyron equation the van t Hoff equation and the Nernst distribution law

all refer to the Gibbs energy changes of relevant processes and they need not be presented as radically different relationships **Bulletin**, 1994 **Contributions to the Data on Theoretical Metallurgy** United States. Bureau of Mines, 1950 **Journal of Research of the National Bureau of Standards** United States. National Bureau of Standards, 1972 *Refrigeration Engineering*, 1938 English abstracts from Kholodil naia tekhnika Multiparameter Equations of State Roland Span, 2000-04-27 The book deals with the most accurate method to describe thermodynamic property data with empirical multiparameter equations of state Due to new theoretical approaches to increasing demands on the accuracy of thermodynamic property data and to increasing computer power such equations became a valuable tool for every day calculations in scientific and engineering applications rather than just the basis of printed property charts and tables The book is dedicated both to users who apply such formulations either in form of commercially available software or in form of programs written by themselves and to scientists engaged in the development of empirical equations of state Starting from a brief history it covers the fundamentals of this subject as well as the most recent developments in the fields of highly accurate reference equations of equations for advanced technical applications and of the description of mixtures with multiparameter equations of state *Refrigerating Engineering*, 1916 Vols 1 17 include Proceedings of the 10th 24th 1914 28 annual meeting of the society *A.S.R.E. Journal* American Society of Refrigerating Engineers, 1921 Entropies of Inorganic Substances Kenneth Keith Kelley, 1950 **Thermodynamics of Geothermal Fluids** Andri Stefánsson, Thomas Driesner, Pascale Bénézech, 2018-12-17 Volume 76 of Reviews in Mineralogy and Geochemistry presents an extended review of the topics conveyed in a short course on Geothermal Fluid Thermodynamics held prior to the 23rd Annual V M Goldschmidt Conference in Florence Italy August 24 25 2013 It covers Thermodynamics of Geothermal Fluids The Molecular Scale Fundament of Geothermal Fluid Thermodynamics Thermodynamics of Aqueous Species at High Temperatures and Pressures Equations of State and Transport Theory Mineral Solubility and Aqueous Speciation Under Hydrothermal Conditions to 300 C The Carbonate System as an Example Thermodynamic Modeling of Fluid Rock Interaction at Mid Crustal to Upper Mantle Conditions Speciation and Transport of Metals and Metalloids in Geological Vapors Solution Calorimetry Under Hydrothermal Conditions Structure and Thermodynamics of Subduction Zone Fluids from Spectroscopic Studies and Thermodynamics of Organic Transformations in Hydrothermal Fluids Thermodynamics Arthur Shavit, Chaim Gutfinger, 2008-12-09 There are many thermodynamics texts on the market yet most provide a presentation that is at a level too high for those new to the field This second edition of Thermodynamics continues to provide an accessible introduction to thermodynamics which maintains an appropriate rigor to prepare newcomers for subsequent more advanced topics The book p **The Planetary Scientist's Companion** Katharina Lodders, Bruce Fegley, 1998 Scientists have collected a wealth of physical and chemical data for the Sun planets and small bodies in our solar system but until now this information has been scattered throughout the technical literature The Planetary Scientist s Companion solves this problem providing for the first

time a single extensive reference for the interdisciplinary fields of planetary science and cosmochemistry The book begins with a summary of frequently used physical and chemical constants unit conversion factors properties of some compounds and minerals thermodynamic data partition coefficients and useful formulas This is followed by an overview of the solar system including comparative data for the planets and their satellites and abundances of the elements Much of the book is devoted to a series of chapters describing in turn the Sun each of the planets and the groups of small bodies asteroids comets meteorites and Kuiper Belt and Centaur objects Each chapter includes an introduction followed by tables of physical and chemical properties compiled from many sources including data on planetary atmospheres surfaces and interiors The book concludes with data on nearby stars the interstellar medium and recently discovered brown dwarfs and possible extrasolar planets followed by a glossary A unique and practical resource for anyone interested in contemporary planetary science and cosmochemistry this volume is likely to be an essential tool in future research Bulletin United States. Bureau of Mines,1949 **National Bureau of Standards Circular** ,1947 **Scientific American** ,1922 Monthly magazine devoted to topics of general scientific interest

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