

TABLE A-1

Molar mass, gas constant, and critical-point properties

Substance	Formula	Molar mass, M kg/kmol	Gas constant, R kJ/kg·K*	Critical-point properties		
				Temperature, K	Pressure, MPa	Volume, m ³ /kmol
Air	—	28.97	0.2870	132.5	3.77	0.0883
Ammonia	NH ₃	17.03	0.4882	405.5	11.28	0.0724
Argon	Ar	39.948	0.2081	151	4.86	0.0749
Benzene	C ₆ H ₆	78.115	0.1064	562	4.92	0.2603
Bromine	Br ₂	159.808	0.0520	584	10.34	0.1355
<i>n</i> -Butane	C ₄ H ₁₀	58.124	0.1430	425.2	3.80	0.2547
Carbon dioxide	CO ₂	44.01	0.1889	304.2	7.39	0.0943
Carbon monoxide	CO	28.011	0.2968	133	3.50	0.0930
Carbon tetrachloride	CCl ₄	153.82	0.05405	556.4	4.56	0.2759
Chlorine	Cl ₂	70.906	0.1173	417	7.71	0.1242
Chloroform	CHCl ₃	119.38	0.06964	536.6	5.47	0.2403
Dichlorodifluoromethane (R-12)	CCl ₂ F ₂	120.91	0.06876	384.7	4.01	0.2179
Dichlorofluoromethane (R-21)	CHCl ₂ F	102.92	0.08078	451.7	5.17	0.1973
Ethane	C ₂ H ₆	30.070	0.2765	305.5	4.48	0.1480
Ethyl alcohol	C ₂ H ₅ OH	46.07	0.1805	516	6.38	0.1673
Ethylene	C ₂ H ₄	28.054	0.2964	282.4	5.12	0.1242
Helium	He	4.003	2.0769	5.3	0.23	0.0578
<i>n</i> -Hexane	C ₆ H ₁₄	86.179	0.09647	507.9	3.03	0.3677
Hydrogen (normal)	H ₂	2.016	4.1240	33.3	1.30	0.0649
Krypton	Kr	83.80	0.09921	209.4	5.50	0.0924
Methane	CH ₄	16.043	0.5182	191.1	4.64	0.0993
Methyl alcohol	CH ₃ OH	32.042	0.2595	513.2	7.95	0.1180
Methyl chloride	CH ₃ Cl	50.488	0.1647	416.3	6.68	0.1430
Neon	Ne	20.183	0.4119	44.5	2.73	0.0417
Nitrogen	N ₂	28.013	0.2968	126.2	3.39	0.0899
Nitrous oxide	N ₂ O	44.013	0.1889	309.7	7.27	0.0961
Oxygen	O ₂	31.999	0.2598	154.8	5.08	0.0780
Propane	C ₃ H ₈	44.097	0.1885	370	4.26	0.1998
Propylene	C ₃ H ₆	42.081	0.1976	365	4.62	0.1810
Sulfur dioxide	SO ₂	64.063	0.1298	430.7	7.88	0.1217
Tetrafluoroethane (R-134a)	CF ₃ CH ₂ F	102.03	0.08149	374.2	4.059	0.1993
Trichlorofluoromethane (R-11)	CCl ₃ F	137.37	0.06052	471.2	4.38	0.2478
Water	H ₂ O	18.015	0.4615	647.1	22.06	0.0560
Xenon	Xe	131.30	0.06332	289.8	5.88	0.1186

*The unit kJ/kg·K is equivalent to MPa·m³/kg·K. The gas constant is calculated from $R = R_u/M$, where $R_u = 8.31447$ kJ/kmol·K and M is the molar mass.

Source of Data: R. A. Kobe and R. E. Lynn, Jr., *Chemical Review* 52 (1953), pp. 117–236; and ASHRAE, *Handbook of Fundamentals* (Atlanta, GA: American Society of Heating, Refrigerating and Air-Conditioning Engineers, Inc., 1993), pp. 16.4 and 36.1.

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Liying Dong



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Thermodynamic Properties of Individual Substances: Calculation of the thermodynamic properties Valentin Petrovich Glushko,1967 **Winter Annual Meeting** American Society of Mechanical Engineers, **Veterinary Anesthetic and Monitoring Equipment** Kristen G. Cooley,Rebecca A. Johnson,2018-10-30 Veterinary Anesthetic and Monitoring Equipment is the first veterinary specific resource solely dedicated to anesthetic and monitoring equipment used in clinical practice Offers a practical guide to anesthetic and monitoring equipment commonly used in veterinary medicine Provides clinically oriented guidance to troubleshooting problems that may occur Discusses general principles applicable to any equipment found in the practice Presents information associated with novel anesthetic equipment and monitors Paper ,1987

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U.S.S.R. Computational Mathematics and Mathematical Physics ,1963 *Journal of Energy Resources Technology* ,1988 *The National Union Catalog, Pre-1956 Imprints* Library of Congress,American Library Association. Committee on

Resources of American Libraries. National Union Catalog Subcommittee,1972 *An Anthropogenic Table of Elements* Timothy Neale,Courtney Addison,Thao Phan,2022-08-31 *An Anthropogenic Table of Elements* provides a contemporary rethinking of Dmitri Mendeleev s periodic table of elements bringing together elemental stories to reflect on everyday life in the Anthropocene Concise and engaging this book provides stories of scale toxicity and temporality that extrapolate on ideas surrounding ethics politics and materiality that are fundamental to this contemporary moment Examining elemental objects and forces including carbon mould cheese ice and viruses the contributors question what elemental forms are still waiting to emerge and what political possibilities of justice and environmental reparation they might usher into the world Bringing together anthropologists historians and media studies scholars this book tests a range of possible ways to tabulate and narrate the elemental as a way to bring into view fresh discussion on material constitutions and thereby new ethical stances responsibilities and power relations In doing so *An Anthropogenic Table of Elements* demonstrates through elementality that even the smallest and humblest stories are capable of powerful effects and vast journeys across time and space **Classed Subject Catalog** Engineering Societies Library,1963 **International Critical Tables of Numerical Data, Physics, Chemistry and Technology: Viscosity and fluidity** National Research Council (U.S.),1929 **Urban Airborne Particulate Matter** Fathi Zereini,Clare L. S. Wiseman,2011-02-16 This book presents the most up to date research and information regarding the origin chemistry fate and health impacts of airborne particulate matter in urban areas a topic which has received a great deal of attention in recent years due to documented relationships between exposure and health effects such as asthma With internationally recognised researchers and academics presenting their work and key concepts and approaches from a variety of disciplines including environmental and analytical chemistry biology toxicology mineralogy and the geosciences this book addresses the topic of urban airborne particulate matter in a comprehensive multidisciplinary manner Topics and research addressed in the book range from common methodological approaches used to sample and analyse the composition of airborne particulates to our knowledge regarding their potential to impact human health and the various policy approaches taken internationally to regulate particulate matter levels **Principles of Chemical Vapor Deposition** D.M. Dobkin,M.K. Zuraw,2013-03-09 *Principles of Chemical Vapor Deposition* provides a simple introduction to heat and mass transfer surface and gas phase chemistry and plasma discharge characteristics In addition the book includes discussions of practical films and reactors to help in the development of better processes and equipment This book will assist workers new to chemical vapor deposition CVD to understand CVD reactors and processes and to comprehend and exploit the literature in the field The book reviews several disparate fields with which many researchers may have only a passing acquaintance such as heat and mass transfer discharge physics and surface chemistry focusing on key issues relevant to CVD The book also examines examples of realistic industrial reactors and processes with simplified analysis to demonstrate how to apply the principles to practical situations The book does not attempt to exhaustively survey the literature or to intimidate

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