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Extraterrestrial Planets

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**Rosina Ehmman**



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web the molar volume will be the volume occupied by one molar mass  $32 \text{ g}$  of the liquid expressing the density in liters instead of  $\text{ml}$  we have  $v = \frac{m}{\rho}$   $32 \text{ g mol}^{-1}$   $790 \text{ g l}^{-1}$   $0.0405 \text{ l mol}^{-1}$  nonnumber

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web the molar volume has the si unit of cubic metres per mole  $\text{m}^3 \text{mol}^{-1}$  although it is more typical to use the units cubic decimetres per mole  $\text{dm}^3 \text{mol}^{-1}$  for gases and cubic centimetres per mole  $\text{cm}^3 \text{mol}^{-1}$  for liquids and solids

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web zero degrees celsius or 273 k is defined as standard temperature and one atmosphere or 760 mm hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure stp is 22.4 l

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