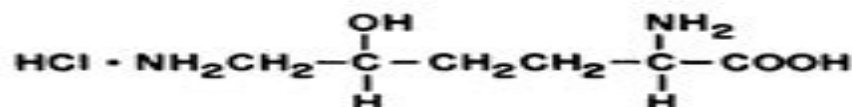


5-Hydroxy-L-lysine MonohydrochlorideFormula: $C_6H_{15}ClN_2O_3$

Formula Wt.: 198.65

Calc. %: C, 36.27; H, 7.61;

Cl, 17.85; N, 14.10; O, 24.16



Source or Method of Preparation: Natural sources (gelatin), or resolution of racemate.

BIBLIOGRAPHY: P. B. Hamilton and R. A. Anderson, *Biochem. Prepn.*, 8, 55 (1961); W. S. Fones, *Biochem. Prepn.*, 8, 62 (1961); W. S. Fones, *J. Am. Chem. Soc.*, 75, 4865 (1953); J. P. Greenstein and M. Winitz, *Chemistry of the Amino Acids*, John Wiley & Sons, Inc., New York (1961), pp. 1996-2016.

Specific Rotation: +17.8° in 6 N HCl, calculated as free 5-hydroxy-L-lysine, $c = 2$, $t = 25^\circ C$. Epimerization lowers the rotation of the product obtained from gelatin hydrolyzates.

Homogeneity: Determined by paper chromatography.

One dimensional: Solvents 1 and 2.

Color Reagent: Ninhydrin.

R_f 0.67, Solvent 1, ascending.

R_f 0.06, Solvent 2, ascending.

R_{BCP} 0.15, Solvent 3a, descending. R_{BCP} refers to the distance moved by the compound calculated as the fraction of the distance moved by bromocresol purple, which is applied as a 0.1% w/v solution in ethanol.

Volatile Matter: Not more than 0.5%.

Water-Insoluble Material: 5 g in 50 ml of water has a turbidity not greater than that given by 0.4 mg of $BaSO_4$ in 50 ml of water.

Ash (sulfated): Less than 0.1%.

Heavy Metals (as Pb): Not more than 20 ppm.

Likely Impurities: 5-*allo*-Hydroxy-D-lysine, 5-*allo*-hydroxy-L-lysine, histidine, lysine, ornithine.

Crystallization Medium: Water, then ethanol to 60% v/v, with further additions of ethanol to 90% as crystallization proceeds.

Stability and Storage: Stable.

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**Society for Experimental Biology and
Medicine (New York, N.Y.)**



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