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web jul 12 2019 when we neutralize a weak acid with a strong base we get a salt that contains the conjugate base of the weak acid this conjugate base is usually a weak base for example sodium acetate NaCH₃CO₂ is a salt formed by the reaction of the weak acid acetic acid with the strong base sodium hydroxide

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web summary an Arrhenius acid produces H⁺ and an Arrhenius base produces OH⁻ in aqueous solutions acids taste sour may sting and neutralize bases bases taste bitter feel slippery and neutralize acids naming acids binary acids contain a single anion HX to name hydro anion with IC end acid

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web definition of acids and bases identifying them along with the conjugate acid and conjugate base pH of strong and weak acids and bases the pH and pOH relationship K_a and K_b pK_a and K_a correlation the acid base properties of salts identifying the acidity of the salt based on the acid and the base it is prepared from and calculating the