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Teacher's guide to Book T2



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School Mathematics Project



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web solution $v_1 = 250 \text{ l}$ $v_2 = 303 \text{ k}$ $t_1 = 273 \text{ k}$ $t_2 = 343 \text{ k}$ we will be applying charles law as pressure is constant here $v_1 t_1 = v_2 t_2$ or $v_2 = v_1 t_1 \times \frac{t_2}{t_1}$ $v_2 = 250 \text{ l} \times \frac{343 \text{ k}}{273 \text{ k}} = 312 \text{ l}$ at 90°C a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml