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Prepared as Background for
**Testing for Effects
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Testing For Effects Of Chemicals On Ecosystems

Scott C. Dulebohn



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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 + 4\text{H}_2\text{O} + 2\text{CH}_4 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O} + 8\text{H}_2 + 3\text{C}_3\text{H}_8 + 5\text{O}_2$

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 $3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} \rightarrow \text{KBr} + \text{CO}_2 + \text{GaBr}_3 + \text{NaBr} + \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{H}_2\text{O} + 3\text{NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} \rightarrow \text{KBr} + \text{CO}_2 + \text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{H}_2\text{O} + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$ using the following equation
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 $3\text{hbr} + 1\text{khco}_3 \rightarrow 1\text{h}_2\text{o} + 1\text{kbr} + 1\text{co}_2$
 $2\text{gabr}_3 + 3\text{na}_2\text{so}_3 \rightarrow 2\text{ga}_2\text{so}_3 + 3\text{nabr}$
 $2\text{sno} + 2\text{nf}_3 \rightarrow 1\text{n}_2\text{o}_3 + 3\text{snf}_2$
using the following equation
 $2\text{naoh} + \text{h}_2\text{so}_4 \rightarrow 2\text{h}_2\text{o} + \text{na}_2\text{so}_4$

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