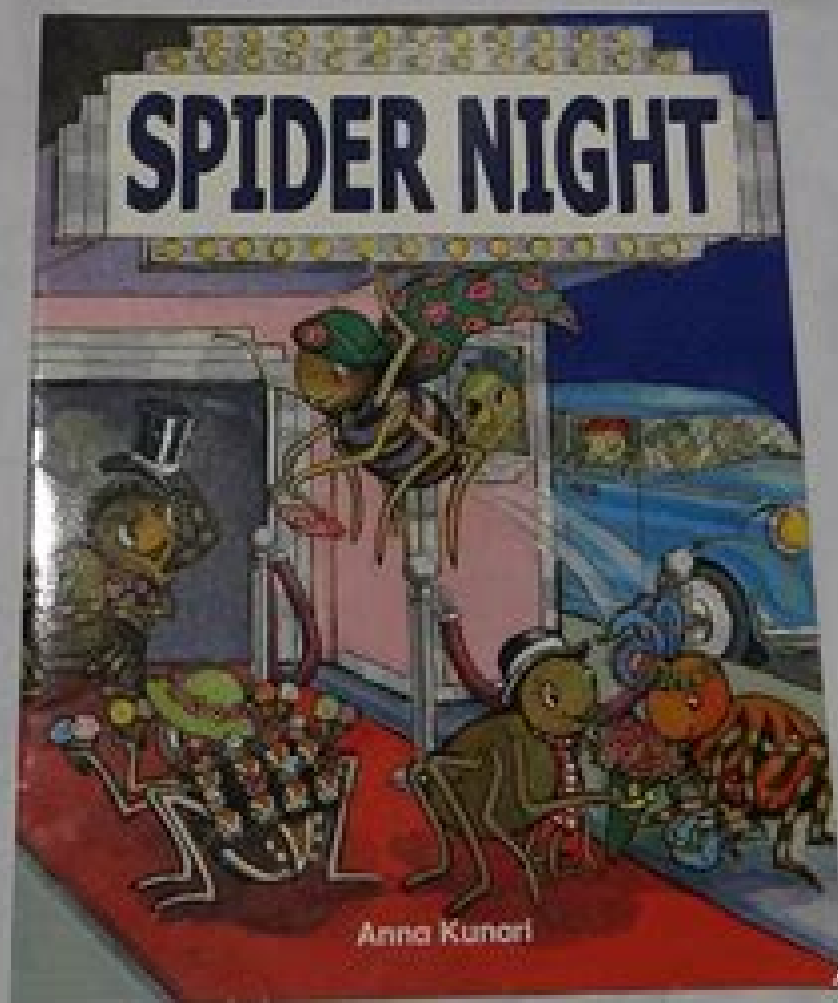


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web balancing chemical equations ks3 gcse answers chemistrytutor me 2018 page 3 of 3 $38 \text{Fe} + 2 \text{O}_3 \rightarrow 3 \text{Co} + 2 \text{Fe}_3\text{O}_4$ $39 \text{H}_2 + 3 \text{PO}_4 \rightarrow 3 \text{BaO} + \text{Ba}_3\text{PO}_4$ $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ $4 \text{Mg} + \text{NH}_4\text{PO}_4 \rightarrow 2 \text{P} + 2 \text{O}_7$ $2 \text{NH}_3 + \text{H}_2 + \text{O}_2 \rightarrow 4 \text{H}_2\text{O} + 2 \text{Al}_2\text{SO}_4$ $3 \text{H}_2 + 2 \text{Al}_2\text{O}_3 \rightarrow 4 \text{H}_2\text{O} + 10 \text{HCl}$ $3 \text{HPO}_3 + 4 \text{HNO}_2 + 2 \text{H}_2\text{O} + 3 \text{O}_2 \rightarrow 4 \text{HNO}_3$

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